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Levels and Distribution of Trace Metals in the Southern Wetlands of Iraq

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Abstract

Trace metals Cd, Co, Cu, Fe, Mn, Ni, Pb and Zn were determined in water and sediments from southern Iraqi wetlands represented by Marshes, (Al-Hawiezh, Central and Al-Hammar), Shatt Al-Arab, Garmat Ali and Northern Arabian Gulf. Subsurface water samples and bottom sediments were collected from each site of sampling on a seasonal basis during the period July/2004 to July/2005. Trace metals were extracted from each sample by means of standard methods and determined by adopting of Atomic Absorption Spectrophotometer. The highest mean concentrations for trace metals in water (in µg/l) were 85.97(Cd) in Beeshah, 407.98(Co) in Qurna, 57.83(Cu) in Mashab, 574.33(Fe) in Khite, 85.34(Mn) in Seebah, 181.0(Ni) in Mashab, 90.25(Pb) in Beeshah, and 702.34(Zn) in Beeshah during different seasons of the study, while the highest mean concentrations of trace metals in sediments (in µg/g) were 35.92(Cd) in Central marsh, 836.46(Co) in Central marsh, 48.69(Cu) in Central marsh, 18435.50(Fe) in Al-Hawiezh marsh, 3648.28(Mn) in Qurna, 486.95(Ni) in Garmat Ali, 5435.5 (Pb) in Qurna and 539.8(Zn) in Qurna. Certain sites along Shatt Al-Arab river receive huge amounts of wastes from municipal, industrial and commercial which might increase the levels of pollutants in water and sediments, as it is found in Beesha at the top north of Arabian Gulf as pollutants in the particulate phase of water are precipitated.

Key Wards: Trace metals, Marshes, Sediments, Waters, N.W. Arabian Gulf

1-Introduction

There are numerous types of pollutants in the aquatic environments, such as organic materials, major ions and heavy metals. These pollutants could be introduce to the aquatic environment as a result of urbanization, industrialization and agricultural activities (Al-Kafaji 1996). Heavy metals are the major anthropogenic contaminant of estuarine and coastal waters. Humans have affected aquatic

trace element cycling through direct discharges into streams and atmospheric emissions, which are deposited at varying distances from the source, and the human activities result in substantial inputs to the environment of Sb, As, Cd, Cr, Cu, Pb, Hg, Mn, Ni, Se, Sn, V, and Zn (Brigham 2000). Many heavy metals, even if present in minute quantities are toxic to plants and animals, thus their release to aquatic environments from either natural anthropogenic sources has an adverse effect upon ecosystem health (Catherine and Gabriel 2000).

Trace metals in aqueous solution seldom frond as free ions but exist as ionic complex utilized by vary of organism in organic and hydrated legends which affected mobility, reactivity and solubility (Wood 1989). They are of environmental interest both as limiting nutrients and as toxicants. Cupper, Zinc and partly Nickel are essential to natural organisms in low concentrations, while they are toxic in higher concentrations. Cadmium and Lead are toxic even at low concentrations and not regarded as being micronutrient.

Trace metals appear in fresh water naturally and in trace concentrations and varied according to kind of sediment (Evans and Engel 1994). Speciation of metals in sediments is sometimes necessary to identify pollution sources, especially to distinguish between mineralized or lattice-hold metals and the more bio-available fraction (Kerstin and Forester 1989).

The aims of this study were firstly to quantify heavy metal concentrations in different classes of water categorized between brackish waters (Al Beesha, Northern Arabian Gulf) to fresh water (Al Hawiezh and central marshes and Iraqi inland waters) and to know if their are any temporal correlation among the sites, secondly to identify the environmental conditions associated with enhanced mobility, and to confirm whether those correlations were significant or not between sites, seasons, and metals.

2-Material and methods

Sites chosen for sampling in this study were two sites of Iraqi Marshlands [(1)Al-Hawiezh and (2)Central marshe], three sites along Shatt Al-Arab river [(3)Al-Qurnah, (4)Al-Seeba and (5)Ras Al-Beesha) and three sites at southern Al-Hammar Marsh [(6)Al-Mashab, (7)Al-Khiet and (8)Garmat Ali] as shown in fig.(1). Samples as subsurface water and surface bottom sediments were collected in a fine whether days on a seasonal basis during the period July/2004 to July/2005.

For water samples, polyethylene bottles prewashed by 0.5N HCl and distilled water were used. Each water sample is thoroughly mixed by shaking, and 100 ml of it is transferred into a glass beaker of 250 ml volume, to which 5 ml of conc. Nitric acid is added and heated to boil till the volume is reduced to about 15-20 ml, by adding conc. Nitric acid in increments of 5 ml till the residue is completely dissolved. The mixture is cooled, transferred and made up to 100 ml using metal free distilled water (limgis 2001).

Van veen grab sampler was used to collect bottom sediment from each site. In the lab, sediments were dried, grinded and extracted according to Sturgeon *et al.* (1982) procedure. The concentrations of trace metals, Cd, Co, Cu, Fe, Mn, Ni, Pb and Zn, were determined by

flame Atomic Absorption Spectrophotometer (Pye Unicum Model SP9) in both water and sediments samples. SPSS program was used for statistical analysis.

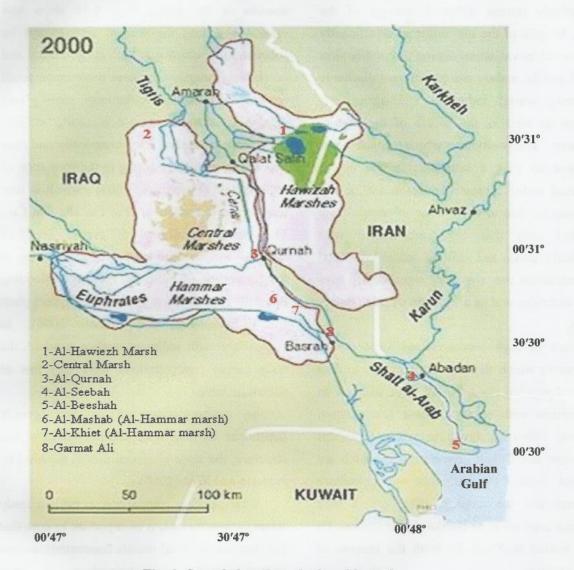


Fig. 1: Sample locations during this study

3-Results and Discussion

The study allows comparing the trace metals concentrations in water and sediments samples from all different sites of Iraq aquatic environments. The distribution and concent-

rations of trace metals Cd, Co, Cu, Fe, Mn, Ni, Pb, and Zn are shown in figs. 2-9 for water and figs. 10-17 for sediment. The highest mean concentrations for trace metals in water (in µg/l) were 85.97(Cd), 407.98(Co), 57.83(Cu), 574.33

(Fe), 85.34(Mn), 181.0(Ni), 90.25(Pb), and 702.34(Zn) at the sites Beesha, Qurna, Mashab, Khiet, Seeba, Mashab, Beesha and Beesha respectively during different seasons of the study. In spite of the low different significantly there are clear variations among the studies sites for all metals, mainly due to different discharge of heavy waste, industrial and agricultural effluent as well as the nature of the bottom sediment compositions which assist that sediment do play a substantial effect in the dissolved metal contents(Anderson 1995), and also may be due to the release of metals from refolded soils that are contaminated with chemicals, mines and military ordinance. Metal concentrations can also vary significantly over short distances and as a function of tide (Batley 1995).

Most of high concentrations were during summer in which the high temperature lead to high evaporation rates, but all were still low in comparison with local standard levels (Iraqi Gazette, 1990) as shown in table (1) except Cadmium and Cobalt concentrations which are resulted from the influence of agriculture effluents and the damage effect of a pipe carrying crud oil to al-Hartha power station, Pb from leaded fuel and Zn from the erosion of ships skeleton.

Bottom sediment is the substrates which adsorbed and accumulated the metals in its natural conditions (Grot et al 1986). The highest mean concentrations for trace metals in sediment (in µg/g dry) were 35.92(Cd), 836.45(Co), 48.69(Cu), 18435.50(Fe), 3548.22 (Mn), 486.95(Ni), 543.75(Pb), and 539.80(Zn)

at central marsh, central marsh, central marsh, Al-Hawieza, Al-Qurna, Garmat Ali, Al-Qurna and Al-Qurna respectively during different seasons of the study. Figs. 9-16 show low variations among the studied sites except Fe concentration which was high at the central and Al-Hawiezh marshes because these areas could be severely burned after drainage, some places with high surface organic covering sulfidic pyrite soils beneath altered chemically and then exposed to oxygen resulting in the formation of sulfuric acid (Fitzpatrick, 2004), which in turn lead to the liberation of Fe, Ca, Mg and Cu. The overall conditions of certain areas of marshes will become hard like ceramic due to the conversion of FeS2 to Fe2O3, on reflooding this soil will not rewet and cannot support plant life (Richardson and Hussain 2006).. In comparison with sediment quality standard, the trace metals concentrations were very low as shown in table (1).

The accumulation effects are greatly dependent on the sediment composition and structure, the same observation was reported by Hussain and Khan(2000).

When making comparison to present study water concentrations, it is important to note that the data are for total metals concentrations and so that be higher when comparison with other studies in Iraq(table 2), but lower than Hussain and Khan(2000) and Guien and Martin(2002) whom are detected total metal concentrations.

Table (1): Mean of trace metals concentrations in water and sediments from southern wetlands of Iraq with mean standard concentrations.

				Water					Sedim	ent	
metal	Present study 2004- 2005	EQS standar ds 2001	EU standards 1998	WHO standards 1993	Water quality Standards 1990	SPA Scope 1980	USEPA 1976+ 1992	Present study 2004-2005	NOAA 2000	CEQG 2002	USEP A 2004
Cd	7.86	10	5	3	5	10	5	13.79	1.2-9.6	0.6-3.5	
Co	265.23	-	1 1 30	-	50	500	-	137.67			
Cu	23.02	40	2000	2000	50	200	1000	12.44	3.4-270	35.7- 197	390
Fe	7621.14	-	200	-	300	2000	100	176.37			
Mn			50	500	100	500	650	28.47			
Ni	154.30	10	20	20	100	200	-	46.55	20.9- 51.6		
Pb	67.62	10	10	10	50	100	50	32.60	46.7- 218	35-91.3	450
Zn	118.51	-	1 - 14-	3000	500	2000	5000	194.96	-150 410	123	410

Table 2: Comparison the present water study trace metals concentration ($\mu g/l$) with the same in other area.

Site	Cd	Co	Cu	Fe	Mn	Ni	Pb	Zn	References
Shatt Al Arab river/Al Iraq	0.26	-	0.90	716	1.30	3.40	0.30	1.80	Abaychi and DouAbul
Shatt Al Arab river/Al Iraq	0.93		0.52		0.07	0.50	0.04	0.80	(1985) Abaychi and Mustafa (1988)
Shatt Al-Arab estuary	0.15	0.33	0.58	389.20	0.69	2.45	0.18	0.93	Al-Saad and Al Kafaj (1996)
Shatt Al-Arab estuary	0.19	0.37	0.47	173.00	1.52	2.85	0.23	0.82	Al-Khafaji (1996)
Shatt Al Hilla	1.11	3.27	1.81	6.74	0.96	0.27	4.21	8.73	Al-Taee (1999)
Karnafully estuary	8-168		125- 482	4200- 22230	282- 931	101-572	33- 540	206-985	Hussain and Khan(2000) Guien and
Danube delta	1.1- 2.4	16-19	201- 1092	3.78- 4.04%	1286- 2290	66-71	58-65	212-218	Martin (2002)
Shatt Al-Arab / Iraq	5.00	50.00	50.00	-	-	100.00	50.00	100.00	Analytical Report
Iraqi wetlands	7.86	265.23	23.02	7621.14	49.44	154.30	67.62	118.51	(2004) Present study

The comparison of sediments average concentrations(μg/g) with other studies inside and outside Iraq (table 3), reveal that Cu, Ni and Mn concentrations were rather low while concentrations of Cd, Co, Zn and Pb were higher, but still low in comparison with sediment quality standard(USEPA, 2004) and slightly higher than NOAA(2000) and CEQG (2002) for Zn and Cd metals concentrations.

According to tables 4, 5 and 6, statistical investigations revealed the presence of positive significant correlations between all studied trace metals except Co in water, while different correlations were found between trace metals concentrations in the sediment and the distribution of trace metals in sediments seems not to be related to the corresponding distribution in water.

Table 3: Comparison the present sediment study trace metals concentrations (µg/g dry) with the same in other area.

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Site	Cd	Co	Cu	Fe	Mn	Ni	Pb	Zn	References
Shatt Al Arab river /Al Iraq	0.18	17.70	44.00	34.70	642.00	646.00	11.30	34.70	Abaychi&DouAbu I(1985)
Al- Kuwait	1.50	0.91	20.5.	280.00	4099.00	96.90	22.70	280.00	Samhan <i>et al</i> (1986)
Shatt Al-Arab river	0.03	17.40	39.60	5.80	914.00	57.20	19.00	5.80	Abaychi&Al- Saad(1988)
Arab Gulf	0.03	18.60	24.20	25.20	751.00	39.80	6.60	25.20	Abaychi&Al- Saad(1988)
Kur Al-Zubier	0.26	-	28.00	72.00	541.00	90.00	29.00	72.00	Al-Edanee et al (1991)
Shatt Al-Arab river	0.05	-	30.00	135.00	917.00	421.00	22.50	135.00	Al-Mudafar et al(1992)
Al-Hammar lake	-	58.00	42.00	3149.00	700.00	197.00	_	88.00	Abaychi (1995)
Shatt Al-Arab estuary	0.27	16.98	29.24	31.99	404.10	104.20	17.74	31.99	Al-Kafaji (1996)
Shatt Al-Arab estuary	-	53.00	36.20	80.40	1063.30	398.60	63.10	80.40	Tarik & Thoraya (1997)
Shatt Al-Hilla	3.92	49.12	34.45	73.41	239.07	193.04	58.20	73.41	Al-Taee (1999)
Gurna	-	42.90	-	388.00	684.00	-	-	388.00	Itawi et al (2000)
Karnafully	0.135-		10118	559.66-	11.6-	16.99-	11-	0.16-	Hussain & Khan
estuary	1.08		-36.25	1117.78	29.56	35.11	36.21	7.47	(2000)
Unpolluted sediment	0.11	-	33.00	95.00	770.00	-	19.00	95.00	GESAMAP,1982
Iraqi wetlands	13.79	137.67	12.44	176.37	28.47	46.55	32.60	194.96	Present study

Tabel(4): Correlations between water trace metals concentrations

	Cd	Ni	Cu	Zn	Fe	Pb	Co	Mn
Cd Pearson Correl	atio 1							
Ni Pearson Correl	atio .812**	1						
Cu Pearson Correl	atid .727**	.568**	1					
Zn Pearson Correl	atio .766**	.815**	.732**	1				**********
Fe Pearson Correl	atid .612**	.668**	.287*	.378*	1			
Pb Pearson Correl	atid .738**	.713**	.535**	.723**	.499**	1		
Co Pearson Correl	atid155	124	029	199	.139	.266	1	
Mn Pearson Correl	atid .497**	.752**	.258	.519**	.388*	.471**	073	

^{**} Correlation is significant at the 0.01 level (2-tailed).

^{*} Correlation is significant at the 0.05 level (2-tailed).

Tabel(5): Correlations between sediments trace metals concentrations

	S. C.	S N S C C C S S C C S S C C S S C C S S C C S S C C S S C C S	(r) = 2	S CC	FES	PB.S	Co.S	S.N.S.	Co.S MN.S Cd,W NI.W	N.IX	.i.≽	N.VZ	ZN.W FE.W PB.W	B.W €	.
CD.SEDIN Pearson Correlati															
ZN.SEDIM Pearson Correlati- 010	-010	~													
NI.SEDIM Pearson Correlati		.609** 064	-												
CU.SEDIM Pearson Correlati	.092	.030	.302	~										-	
FE.SEDIM Pearson Correlati		.526**- 120	.638*	.357*	1										
PB.SEDIM Pearson Correlati	.012	174	.235	.027	019	-									
CO.SEDIN Pearson Correlatif. 109 -	-,109	010	117	304	.035117	-,117	_								
MN.SEDIN Pearson Correlati 084	.084	560*	560*1.016	150	.245	045	- 004	1							
CD.WATEI Pearson Correlatil .123	.123	221	.146	138	.267	.267012	197	052	-						
NI.WATER Pearson Correlati	.235	-,163		138	.337*	.337* .019	.337*018	018	.812*					8	
CU WATEI Pearson Correlatii. 060	090	.232	029	148	.123	.102	168	.469**	727*	.568*1	-				
ZN WATEF Pearson Correlati	249	145	890.	.029	.275	000	.239	.321	.766*	.815**	.732**	~			
FF WATER Pearson Correlati	167	-322	.208	.105	.297	155	.313	-165	.612*	.668*1	.287	378*	-		1
PB WATER Pearson Correlati 118	, C	+	- 053		- 132	258	.323	- 183	738*	.713**	.535**	.723**	499*	-	
CO WATEI Pearson Correlatil, 233					344*	1	1.172	-,133	155	- 155 - 124 - 029 - 199	029	199	.139	.266	-
A HOLONDO CONTRACTOR AND			104		180	1	278	- 164		497* 752*	.258	.519*	388	471*1.073	3

**. Correlation is significant at the 0.01 level (2-tailed).

". Correlation is significant at the 0.05 level (2-tailed).

Tabel(6):Correlations between water and sediments trace metals concentrations

	Cd	Ni	Cu	Zn	Fe	Pb	Co	Mn
Cd Pearson Correlation	1							
Ni Pearson Correlation	.812**	1						
Cu Pearson Correlation	.727**	.568**	1					
Zn Pearson Correlation	.766**	.815**	.732**	1				
Fe Pearson Correlation	.612**	.668**	.287*	.378*	1			
Pb Pearson Correlation	.738**	.713**	.535**	.723**	.499**	1		
Co Pearson Correlation	155	124	029	199	.139	.266	1	
Mn Pearson Correlation	.497**	.752**	.258	.519**	.388*	.471**	073	

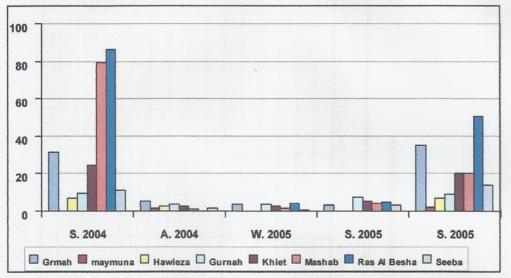


Fig. 2: Seasnal average conc. of Cadmium(Cd)in water (µg/L) from southern wetlands of Iraq

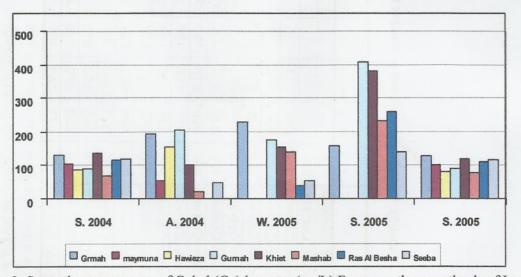


Fig. 3: Seasnal average conc. of Cobalt(Co) in water (µg/L) From southern wetlands of Iraq

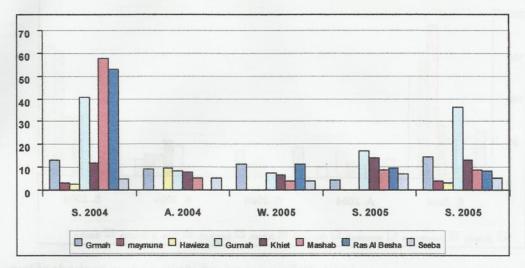


Fig.4: Seasnal average conc. of Copper (Cu)in water (µg/L)from southern wetlands of Iraq

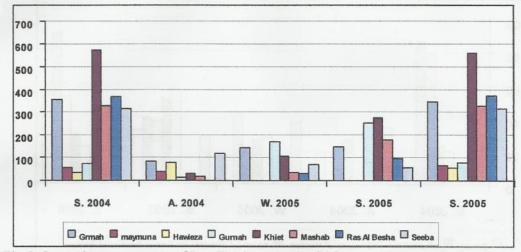


Fig.5: Seasnal average conc. of Iron(Fe) in water (µg/L)from southern wetlands of Iraq

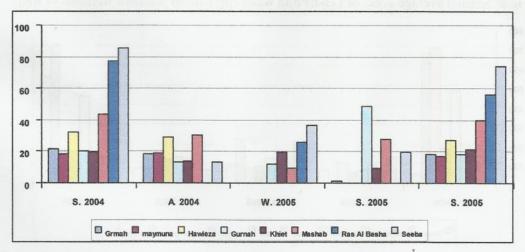


Fig.6 Seasnal average conc. of Manganes (Mn)in water (µg/L)from southern wetlands of Iraq

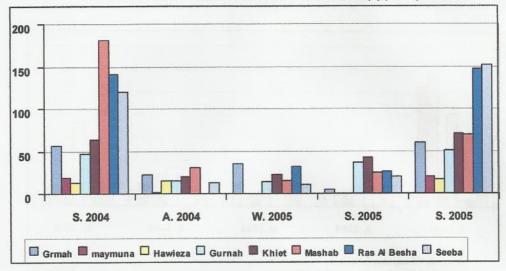


Fig.7: Seasnal average conc. of Nickel(Ni) in water (µg/L)from southern wetlands of Iraq

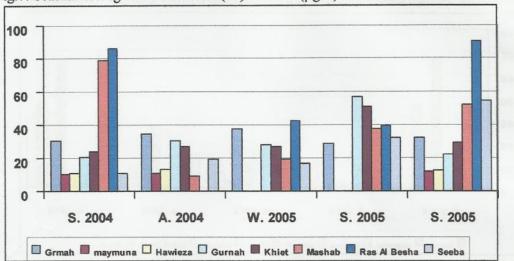


Fig.8: Seasnal average conc. of Lead(Pb)in water (µg/L)from southern wetlands of Iraq

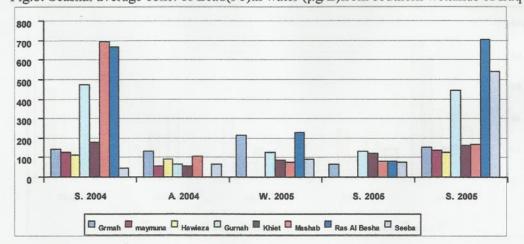


Fig.9: Seasnal average conc. of Zink(Zn) in water (µg/L) from southern wetlands of Iraq

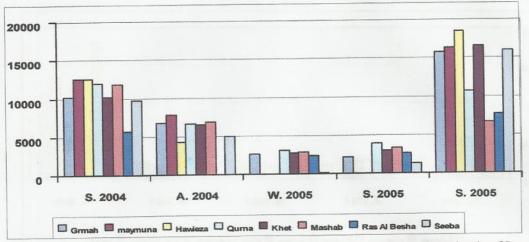


Fig.13: Seasnal average conc. of Iron (Fe)in sediment (µg/gm) from southern wetlands of Iraq

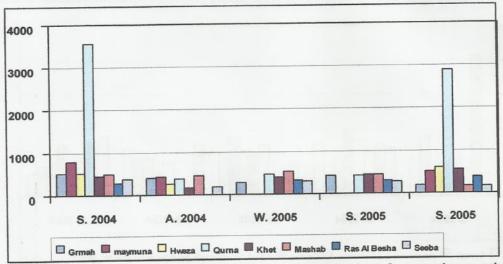


Fig.14 Seasnal average conc. of Manganes (Mn)in sediment (µg/gm) from southern wetlands of Iraq

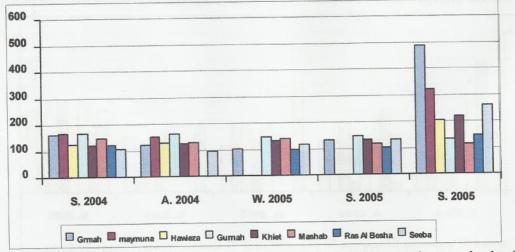


Fig.15: Seasnal average conc. of Nickel(Ni) in sediment (µg/gm)from southern wetlands of Iraq

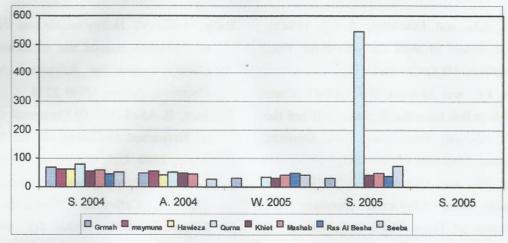


Fig.16: Seasnal average conc. of Lead(Pb)in sediment (µg/gm)from southern wetlands of Iraq

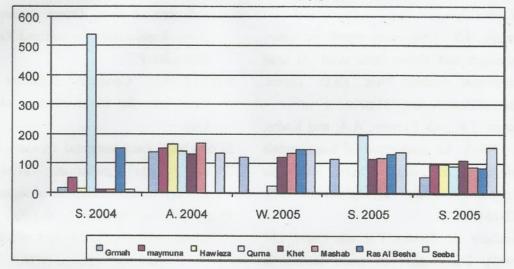


Fig.17: Seasnal average conc. of Zink(Zn) in sediment (µg/gm)from southern wetlands of Iraq

Conclusion

Data reported in this study show variable trends and unclear variations during seasons of study. This could be explain on the basis of alternative inflow of water into dried marshes as well as other sources of effluent, domestic, industrial and atmospheric fall out. Moreover, certain areas contain compounds ready for oxidation by oxygen; this will lead to incorporate the oxidation-reduction mechanisms.

4-References

Anderson, A.G.1995. Heavy metals and phosphates in the Coeur d'Alene Lake. In: Environmental Sampling, Barman, Edinger and Van der Sluvs, (Eds.). Boston, P: 1-56.

Abaychi, J.K. (1995). Trace element distribution and sedimentation rate in Al-Hammar Lake, southern Iraq. Marina Mesopotamica. 10(2):379-392.

- Abaychi, J.K. and DouAbul, A.A.Z. (1985). Trace metals in shatt Al-Arab river, Iraq. Water Res. 19(4):457-462.
- Abaychi, J.K. and Al-Saad, H.T. (1988). Trace metals in fish from the Arabian Gulf and the shatt Al-Arab river. Iraq. Bull. Environ. Contam. Toxically. 40:226-232.
- Abaychi, J.K. and Mustafa, Y.Z. (1988). The Asiatic clam Corbicula fluminea: An Indicator of Trace Metal Pollution in the shatt Al-Arab river. Iraq. Environ. Pollute. 54:109-122.
- Al-Khafaji, B.Y. 1996. Trace metals in water, sediments and fishes from shatt Al-Arab North-west Arabian Gulf. Ph.D. Thesis. Univ. of Basrah, Iraq. 131 p.
- Al Edanee. T.E.; Al- Kareem, A.A. and Kadm, A. (1991). An assessment of trace metals pollution in the Khor Al-Zubair environmental. Mar. Mesopotamica. 6(2): 143-154.
- Al Muddafar, A.; Jassim, T.E. and Omer, I.N. (1992). Distribution of trace metals in sediment and biota from Shatt Al-Arab river Iraq. Mar. Mesopotamica. 7:49-61.
- Al-Saad, H. T. and Al-Kafaji, B.Y. (1996). Distribution of trace metals in water and biota samples from Shatt Al-Arab estuary. Marine Mesopotamica. 11(1):63-77.
- Al Taee, M.S. (1999). Some metals in water, sediments, fishes and plants of the Shatt Al-Arab river. MS.C. Thesis, Univ. of Basrah, Iraq.129.
- Analytical Report No.3201-798/2004. IRAQ Project of CPNS IRAQ, Geo test Brano, TCH.

- Batly, G. E.1995. Heavy metals and tributyl tin in Australian coastal and estuarine waters. CSIRO Center for Advanced Analytical Chemistry. Menai. NSW 2234.
- Brigham, E. Mark. (2000). Elemental chemistry of Streambed Sediments of the St. Croix Basin, 2000. U.S. Geological survey water-Resources Investigations Report 02-4087.

 36p Catherine J. Souch and Gabriel Filippeili. 2000. Water for our future projects. Contamination of wetland sediments Indiana Dunes National Lace shore: Implications for wetland Restoration strategies. P 7
- CEQG.(2002). Canadian Sediment Quality Guidelines for the Protection of Aquatic Life.—
- EQS (2001). Environmental Quality Standards for water pollution. Ministry of Environment, Government of Japan
- Evans. D. and Engle, D. W.1994. Mercury bioaccumulation in fin fish and shell fish from laraca loay. Texas. NOAA Technical memorandum.p.89.
- Fitzpatrick R.W. (2004). Changes in soil and water characteristics of some natural.

 Drained and reflooded soils in the Mesopotamian Marshlands: Implications for Land Management Planning. CSIRO and Water Client Report, N. P.: CSIRO.
- GESAMP (1982). The health of the Oceans. Rep., Stud, GESAMP (15):108 P and UNEP Req. Seas Rep. Stud. (16):108 P, In: Al-Khafaji, B.Y. (1996) Trace metals in water, sediments and fishes from Shatt Al-Arab,

- North-west Arabian Gulf. Ph.D. Thesis. Univ. of Basrah, Iraq.131 p.
- Grot, A.J.;K.H. Zschuppe; W.Salomane 1986.
 Standardization of analysis for heavy metals in sediments. Hydrobiology 92. 689-695.
- Guien, C.; Martin, J. M. (2002). The level and Fate of metals in the Danube River Plume. Estuarine, Coastal and Shelf Scince 54, 501-512.
- Hussain; Md. S. and Khan; S. A. (2000). An Environment Assessment of Metal Accumulation in the Karnafully Estuary, Bangladesh. Chetagong -4331. 115 127.
- Iraqi Gazette. (1980). Instructions No.80406 issued under the maintenance of public water and the rivers from pollution. no.2763 in 31/3/1980 and no. 2766 in 28/7/1980.Iraq.
- Itawi, R.K.; Sabri, A.W.; Al-Jobri, S. M.; and Khalaf, K. (2000). Study of elemental distribution in fish and plant samples from Tigris river using INAA technique. Iraqi. J. of Chem. 26(1). 203-212.
- Leon P. Zann. 2006. Stat of the Marine Environment Report for Australia: Pollution - Technical Annex 2. p.15.
- limgis (2001). Physico chemical and biological analyses of water.
- Richardson, Curtis J. and Hussein N. A.2006.

 Restoring the Garden of Eden: An Ecological Assessment of the Marshes of Iraq. Bioscience. 56.(6): 477-489
- Riley, J.P. and Taylor, D. 1968. Chelating for the concentration of trace elements from sea water and their analytical use in conjunction

- with atomic absorption spectrophotometer. Anal. chem. Acta.40: 479-485.
- Samhan, O.; Zarba, M. and Anderlini, V. (1986). Aeolian contribution of trace metals to marine sediment of Kuwait. Environ. Inter. 12:611-617.
- Scop (1980). Techinipetral . Book no.48002 SPA. U.S. Environmental Protection Agency.(2006). Contaminated Sediment in water
- Sturgeon, R.E.; Desoudneris, J.A.; Berman, S.S. and Russell, D, S. 1982. Determination of trace metals in estuarine sediment by graphite furnace atomic absorption spectrophotometer. Analy. Chem. Acta. 134: 288-291.
- Tarik, E. J. and Thorya, M. A. (1997). An assessment of trace metals pollution in the Shatt Al-Arab river, Iraq. J. Basrah Research, vol.19, part.1. P53-62.
- U.S. EPA. (1976). (U. S. Environmental Protection Agency). Current Drinking Water Standards.
- U.S. EPA. (1992). Current Drinking Water Standards: National Primary Drinking water Regulation, 816-F-013
- U.S. Environmental Protection Agency. (2004).

 Contaminated Sediment in water.
- Wood J. M., 1989. Transport, bioaccumulation and toxicity of elements in micro organisms under environmental stress.7th international conference on Heavy metals in Environment.(Ed. J. P. vernal; CEP consultants Ltd, Edinburgh, U.K. 1. pp.5-9.

مستويات وتوزيع المعادن النزرة في مياه مسطحات العراق الجنوبية

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الملخص

أجريت دراسة لتقدير المعادن النزرة: الكادميوم والكوبات والنحاس والحديد والمنغنيز والنيكل والرصاص والخارصين في مياه مسطحات العراق الجنوبية والمتمثلة باهوا ر الحويزة والوسطي والحمار وشط العرب وكرمة علي وراس البيشة عند شمال الخليج العربي. تم جمع عينات مياه تحت سطحية ورواسب قاعية لقترات موسمية خلال الفترة تموز 2004 إلى تموز 2005. استخلصت المعادن النزرة لكل عينة وتم تقدير تراكيزها بالطرق القياسية وباستخدام جهاز الامتصاص الذري اللهبي. كانت معدلات التركيز العليا للمعادن النزرة في المياه (بوحدات مايكرو غرامالتر)بمقدار 85.95 (كادميوم) في البيشة، 40.79 (كوبلت) في القرنة، 57.83 (نحاس) في المسحب ، 574.33 (مديد) في البيشة. و 574.34 (حديد) في البيشة. و 574.34 (خوسطي و مختلفة، بينما كانت أعلى معدلات للتركيز في الرواسب (بوحدات مايكوغراماغم وزن جاف)بمقدار 35.95 ((كوبلت) في الهور الوسطي و 36.48 (نحاس) في القرنة و 3648.28 (نحاس) في القرنة و 3648.28 (نحاس) في القرنة و 3648.24 (حديد) في الحويزة و 3648.28 (نيكل) في القرنة و وهور الحويزة بأنها تمنقبل كميات كبيرة من مخلفات منزلية وصناعية وحركة نقل بالزوارق حيث سحلت مستويات عليا من الملوثات

كلمات دالة: معادن نزرة، اهوار، رواسب، مياه، شمال غرب الخليج العربي.