# Irradiation Assisted Synthesis and Antifungal Activities of Some Mannich Bases\*

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#### **ABSTRACT**

To eliminate the use of solvent during the course of reaction and to decrease the reaction time in addition to enhanced the yield in synthesis of some Mannich bases 5-((5-(Aryloxymethyl)-1,3,4-thiadiazol-2-ylamino)(phenyl)methyl)pyrimidine-2,4,6 (1H,3H,5H)-trione (2a-h),the reaction was carried out under microwave irradiation of equimolecular mixture of 2-amino-5-aryloxymethyl – 1,3,4 –thiadiazoles (1a-h), benzaldehyde and barbituric acid, using magnesium sulfate as absorption support.

In conventional method the attempts to react amine(1a) with aldehyde to form Schiff base and the reaction of this base with barbituric acid did not succeed.

The structures of these compounds were confirmed by IR Spectrophotometer. All these bases were screened for their antifungal activity against *Aspergillus flavus* and found to possess weak activity.

Key Words: Mannich base; Microwave synthesis

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.Aspergillus flavus

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## **INTRODUCTION**

At the beginning of this century, List and Barbas showed that the direct asymmetric Mannich reactions of a broad range of ketones and aldehydes were catalyzed by proline,  $\beta$ -Amino ketones and aldehydes can be synthesized stereoselectively in good yields (Barbas *et al.*, 2003; Bahmanyar and Houk, 2003).

In the same way Kidwai et al described the use of montmorillonite clay under microwaves to synthesis novel Mannich bases of thiobarbiturates and barbiturates (Kidwai *et al.*, 2005).

As part of continuous program directed toward the studies with polyfunctionally substituted heterocyclic's (Shandala *et al.*, 1998, 2001 and 2002; Ayoub *et al.*, 2001; Basher,2005;Al-Gwady, 2007; Saied, 2007), it was became of interest to investigate preparative routs to synthesis of these Mannich bases which contain more than one center of anticipated antifungal activity such as aryloxy heterocyclic ether linkage (Saied, 1983), pyrimidine moiety (Kidwai *et al.*, 2005) and 1, 3, 4-Thiadiazole ring. Several compounds containing one or both of the first two radicals were proved to show promising growth inhibitory action on different fungus especially *Aspergillus flavus* (Kidwai *et al.*, 2005). The insertion of 1,3,4-thiadiazole ring was based on the reported fungicidal activity of some compounds containing this ring (Saied, 1983).

#### **EXPERIMENTAL**

All melting points were determined on a Gallen Kamp and Electro thermal 1A9300 Digital-Series (1998) apparatus and were uncorrected. The IR–spectra (vmax cm<sup>-1</sup>) were recorded on Perkin–Elmer 590B Spectrophotometer. UV-On Shimadzu UV-160 spectrophotometer using EtOH as solvent.

## 2-Amino-5-aryloxymethyl – 1,3,4 – thiadiazoles (1a-h):

All these amines were synthesized according to published procedures and had the same physical properties (Saied, 1983):

General Procedure for the Synthesis of 5-((5-(Aryloxymethyl)-1,3,4-thiadiazol-2-ylamino) (phenylmethyl) pyrimidine-2,4,6 (1H,3H,5H)-trione (2a-h) (Kidwai *et al.*, 2005):

Equimolar mixture of 1a-h (0.01 mol), benzaldehydes (1.06 gm, 0.01 mol) and barbituric acid (1.28gm, 0.01 mole) were mixed with a spatula for few minutes. The reaction mixture was placed in an open conical flask in a domestic microwave oven (containing crucible of 2gm magnesium sulfate) and irradiated for 2 min. (the reaction was TLC monitored to indicate the disappearance of the starting material), on completion the mixture was cooled at room temperature and the product was extracted using ethanol.

Evaporation of the solvent under reduced pressure yielded the corresponding title compounds 2a-h, which were recrystallized from methanol. %Yields, melting points, in addition to IR were listed in Table (1).

# Attempted to synthesis of (2a) by conventional method: (Ralhan et al., 1960)

An equimolar amount of (1a) and benzaldehyde (0.01 mole) were put in a round-bottomed flask. To this, glacial acetic acid in ethanol was added and the reaction mixture was refluxed for 10-12hours with stirrer equipped with condenser. No reaction occurred

as checked with TLC, and the precipitate has the same melting point of starting material (1a).

Table 1: %Yields, melting points and IR of compounds (2a-h)

	Ar	Yield %	Mp °C	IR, KBr ,Cm -1			
Compd. No.2				C=O Stretching	N-H Stretching	C-O-C Sym./ asym.	C=N Cyclic
a	C <sub>6</sub> H <sub>5</sub> -	60	390-91	1650	3340	1190	1690
b	4-ClC <sub>6</sub> H <sub>4</sub> -	55	290-93	1650	3320	1185	1680
С	2,4-Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> -	35	290-91	1640	3390	1210	1695
d	$2,4-Br_2C_6H_3-$	65	310-11	1650	3320	1200	1685
e	4-FC <sub>6</sub> H <sub>4</sub> -	60	331-2	1660	3320	1220	1680
f	4-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -	36	319-20	1640	3340	1195	1687
g	2-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -	30	310-11	1640	3335	1180	1683
h	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	70	315-6	1650	3330	1205	1690

## **FUNGICIDAL EVALUATION**

The colony diameter (mm) and %Inhibition of *Apergillus flavus* on Sabauraud Agar Medium (SAM) treated with 2% compounds (2a-h) were shown in Table (2).

Table 2: Colony Diameter (mm) and %Inhibition of *Apergillus flavus* On (SAM) medium treated with compounds(2a-h)

Compounds	%	Colony Diameter
No.	Inhibition	(mm)
2a	0	90
2b	22.22	70
2c	22.22	70
2d	17.11	74
2e	11.11	80
2f	10.0	81
2g	16.66	75
2h	11.11	80

## **REULTS AND DISCUSSON**

Scheme (1) summarizes the synthetic route followed for the preparation of the designed compounds. In this rout benzaldehyde and 2-amino-5-aryloxymethyl-1,3,4-

thiadiazoles (1a-h) were reacted with barbituric acid under MW irradiation to give Mannich products (2a-h). In this work, the organic compounds adsorbed on the surface of inorganic magnesium sulfate to support the absorption of microwaves irradiations. The bulk temperature was relatively low in such solvent free reactions although highly localized temperatures may be reached during microwave irradiation.

5-((5-(Aryloxymethyl)-1,3,4-thiadiazol-2-ylamino) (phenylmethyl)pyrimidine-2,4,6(1*H*,3*H*,5*H*)-trione

## Scheme (1)

In conventional method the attempts to react amine (1a) with aldehyde to form Schiff base and the reaction of this base with barbituric acid did not succeed as was referred in literatures (Belinelo *et al.*, 2002; kidawi *et al.*, 2005).

All these bases were screened for their antifungal activity against *Aspergillus flavus* and found to possess bad activity. (Table 2).

## **CONCLUSION**

The advantages of this work were the safe and simplicity with a good product yield, short reaction time and exclude the volatile and toxic solvents.

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