

The Effect of Water Film on Built Oxide Layer in Copper Plate Using Laser Shock Wave

Ali A. Abdulhadi*

Received on: 12/ 6/2011

Accepted on: 8/9 /2011

Abstract

After Nd-Yag laser treatment with and without water thin film, copper plate material was used under 2880 hours of humidity environment to build a green oxide layer. Oxide layer and water thin film were studied; corrosion rate and oxide percent were measured with and without water film under same environment for one laser pulse, 2 pulses, 3 pulses and 4 pulses. The results show improvement and stability in corrosion rate for water thin film of one pulse during the test period. This paper gives comparative results referring to water laser technology on corrosion rate of green oxide layer for copper plate with and without water thin film during specific period of time.

تأثير طبقة الماء على تكون طبقة الأوكسيد في صفائح النحاس باستخدام موجة الصدمة بالليزر

الخلاصة

بعد المعاملة بNd-Yag الليزر مع طبقة الماء الرقيقة وبدونها، تم تعريض صفائح النحاس لمدة 2880 ساعة من اجواء الرطوبة لبناء طبقة أكسيد الخضراء. تمت دراسة طبقة الأوكسيد وطبقة الماء الرقيقة؛ تم قياس معدل التآكل ونسبة أكسيد لنفس الظروف بوجود طبقة الماء وبدونها ومن واحد الى اربع نبضات باستخدام الليزر ومقارنة النتائج. أظهرت النتائج تحسنا واستقرار في معدل التآكل في وجود طبقة الماء الرقيقة ولنضضة واحده خلال فترة الاختبار. هذا البحث يعطي نتائج مقارنه في اشارة الى تكنولوجيا الليزر والماء على معدل نمو طبقة الأوكسيد الاخضر لصفائح نحاسية مع وبدون طبقة الماء الرقيقة خلال فترة محددة من الزمن.

Introduction

Copper is a metal that has a wide range of applications due to its good properties. It is used in electronics for production of wires, sheets, tubes, and also to form alloys. Copper is resistant toward the influence of atmosphere and many chemicals, however, it is known that in aggressive media it is susceptible to corrosion. The use of copper corrosion inhibitors in such conditions is necessary since no

protective passive layer can be expected. The possibility of the copper corrosion prevention has attracted from many researchers so until now numerous possible inhibitors have been investigated amongst them there are inorganic inhibitors [1]. At ordinary temperatures, most engineering metals are protected by very thin oxide films of the order 3 to 10 nm thick. These films form very rapidly on contact with atmospheric oxygen subsequent growth in uncontaminated air with low humidity

is usually imperceptible, for this reason that aluminum, chromium, zinc, nickel, and some other common metals remain bright in unpolluted indoor atmospheres [2]. A general corrosion attack on copper is most often associated with soft and acidic waters. It usually proceeds at a slow rate and is characterized by a build-up of cupric acids. The most important factors influencing the general corrosion rate of copper are the pH of the water, softness, temperature, oxygen content of the water and stress (Stress Corrosion). Water that is soft--less than 60 mg/l of hardness--with a pH of less than 6.5 will be aggressive to copper. If the water is heated, the aggressive nature will be greater due to the destruction of the metal-oxide layer on the copper [3] [4].

Moist, oxygen and temperature have most important effect on the material corrosion; moist air is usually insufficient for corrosion to occur. Materials on the surface of metals such as corrosion products and salt deposits can be hygroscopic and can absorb sufficient amounts of water from moist air to allow corrosion to proceed [5].

The laser beam can be used to generate surface acoustic waves in the substrate, where the mechanical forces associated with the wave were sufficient to allow for surface damage to the substrate if the laser beam was not controlled [6].

Laser shock wave is a process that imparts compressive stresses on the surface of a material. It is used primarily for increasing fatigue life and component for stress corrosion cracking [7]. tensile stress is known to be necessary component for stress

corrosion cracking (SCC)[8], whereas compression stress can mitigate SCC initiation or decelerate propagation [9].

Experimental Part

Thermo-hygrometer model 303 was used to measure the natural environment of humidity during corrosion time table (1).2880 hours was the humidity exposure time for all specimens.

Q switched (1.6J), 3×10^{-9} pulse duration Nd-Yag laser system with thin water film was used to generate shock wave. A lens has focal length 10cm was used to irradiate the copper plate B152 (1.5cm x 1.5cm x 0.5mm) with (1.5mm) laser spot diameter. The laser beam was focused on wet copper surface below the transparent solution (distilled water). The liquid height ~ 1mm figure (2).The plasma formation was (~ 2µm) equation (1) [10].

$$L_p (\Delta\tau_p) = 2 \times 10^4 (P \Delta\tau_p / z) \dots \dots \dots (1)$$

P in k bars, z is the shock impedance in g/cm² s, L_p thickness of the interface in µm, and τ_p pulse duration in ns

The interface pressure of plasma formation was (213x10⁸ Pa) equation (2)

$$P_0 = 0.10 \times \sqrt{\left[\frac{fE}{3}\right] Z I_0} \dots \dots \dots (2)$$

I₀ in W/cm²

fE is constant fraction of internal energy ≈ 0.1 – 0.2

Percent of oxide was measured using grid square technique (GST) figure (3).

$$\text{Area} = [(\text{full cells} + 1/2(\text{partial cells})) \times (\text{cell area value})] \dots \dots \dots (3)$$

With the technique described above it will be possible to make estimates of area.

Percent of copper oxide to the total surface was measured starting from 570 hours ending by 2880 hours figures (4 to 7).

The corrosion rate is the corrosion effect on a metal per unit time [11]; the corrosion effect may vary with time and may not be the same at all points of the corroding surface. Therefore, reports of corrosion rates should be accompanied by information on the type, time dependency, and location of the corrosion effect will calculated from the equation (4) [12] at each time intervals (570 ,1140,1710,2280 ,2880) hours.

$$\text{mpy} = 22.273 W_L / (D * A * t) \dots\dots\dots (4)$$

mpy=corrosion rate (mils per year penetration) g, W_L =weight loss in g, D, density g/cm³, A=area in cm², t=time

The corrosion rate of oxide layer was measured after laser treatment with water film and without water film figures (8 to 11).

The laser-shock wave supported by water film method based on the photomechanical effect of a laser pulse has been used to produces a cylindrical convergent surface acoustic wave under ($4.7 * 10^8$ W/cm²) laser intensity, Generating shock waves on the surface will leads to acoustic wave at the surface and inside the surface, wide-band microphone with frequency of 20Hz-20 kHz placing 5cm near to the laser spot diameter, the acoustic wave emission under water thin film was 15720 hertz figure (1) which can enhance to change the surface properties figures (13,14).

Results and Discussion

We have compared the experimental data of the oxide formation percent with

and without water film in laser shock wave processing.

This comparing shows improvement in most specimens of water film during the test period, this improving in corrosion rate can be seen very clearly in figures (8, 9, 10, and 11), figure (9) shows the best improvement under one pulse treatment because of efficient amount and distribution of compressive stress due to laser shot, figures (9, 10 and 11) the oxide rate is become not stable and goose height this due to compressive stress generating beyond the specific limits on the surface, The improvement can be clarify to formation of compressive stresses on the surface this will effect directly of the on the formation of localized corrosion. The localized corrosion leads to oxygen concentration penetrate on the surface and this will be starting point of corrosion figure (11). The stability in corrosion rate can be seen for one pulse, 2 pulses with and without water film in range of first three points of time this happened when exposed to the atmosphere over long periods of time, copper will form coloration on the surface known as patina. When first formed, the patina exhibits a dark color that gradually turns green. The length of time required to form the patina depends upon the atmosphere.

Figures (4, 5, 6 and 7) shows Percent of oxide formation after laser treatment, in figure (6) the percent of oxide at first 570 hours was zero percent but its goes up again in figure (7), this fluctuating is due to compressive stress formation, when it's certain limit its will work on accelerate the penetration of localized

corrosion, below this limit compressive stresses will decelerate the localize corrosion hence oxide formation it can be seen in all periods of test.

In figures (13, 14) shows the acoustic waves emitted from an oxidized copper substrate under the laser irradiation on the same spot. When the first laser pulse is irradiated on the surface, a very strong acoustic wave is observed. This strong acoustic wave implies that the shock wave was generated on the surface and the laser pulse interacted strongly with the oxides surface due to the short laser pulse length. The emission of the acoustic waves is strong; especially in the initial period and decays rapidly due to change in material properties which will affect further interactions.

It is clear that strong absorption and interaction between the laser and the surface to the generation of plasma and the changing the stresses on the surface when water film is found because this will leads to confine the generating waves and reflected directly to the surface this will create strong hammer on the work surface, these waves will transform the tensile stresses in to compressive stresses.

Conclusions.

- The corrosion rate and percent of oxide formation during specific time can be controlled by surface treatment.
- The oxide formation cannot be reach to zero because of the environment effect.
- The shock wave process with water film more efficiently to decelerate the corrosion.

- This type of process can be use efficiently to small devices.

References

- [1]-M. Antonijevic & M Petrovic "Copper Corrosion Inhibitors", Int. J. Electrochem. Sci., 3 1 – 28 (2008).
- [2]-D.Talbott, "corrosion science and technology" CRC Press 1998.
- [3]-P.Yu, M. Liao, "corrosion" 59(2003)340.
- [4]-P.Schweitzer, "fundamental of metallic corrosion", CRC Press, 2007.
- [5]- "Corrosion Control" NAVFAC MO307 September 1992.
- [6]- A.A. Kolomenskii, H.A. Schuessler, V.G. Mikhalevich and A.A. Maznev, *J. Appl. Phys.* 84,2404 (1998).
- [7]- E. Kannatey, "Principles of Laser Material Processing", Willy. (2009).
- [8] M.O. Speidel, "Metallurgical and Materials Transactions" A 6A (1975) 631–651.
- [9] R.H. Jones, "Stress corrosion cracking of aluminum alloys", in: R.H. Jones (Ed.), Stress-Corrosion Cracking.
- [10]- E. Kannatey, "Principles of Laser Material Processing", Willy. (2009).
- [11]-M. Hernández, A. Juárez, R. Hernández "Interferometric thickness determination of thin metallic films, Superficies y Vacío", 9, 283-285, December, 1999.
- [12]-R.Kelly, "Electrochemical Techniques in Corrosion Science and Engineering", Marcel Dekker, (2003).

Table (1) Shows the Humidity Percent for January, February, March and April.

Humidity (%)			Humidity (%)			Humidity (%)			Humidity (%)		
high	avg	low	high	avg	low	high	avg	low	high	avg	low
93	74	45	77	58	46	94	75	49	58	34	21
87	65	46	93	68	46	100	80	49	45	30	15
88	67	35	72	51	22	93	59	40	59	34	15
94	77	52	71	49	27	82	54	28	52	27	9
100	75	33	87	51	24	93	64	43	57	31	13
87	62	34	93	80	40	67	44	23	43	21	12
76	49	28	81	61	36	59	39	20	64	38	21
76	57	37	81	56	31	72	47	28	68	43	26
87	58	31	66	47	33	59	38	18	52	28	15
66	42	22	66	46	28	41	28	15	48	29	16
81	48	27	66	46	26	60	37	20	45	25	17
66	44	23	76	47	23	49	34	19	57	34	20
59	49	40	47	39	27	56	33	13	78	43	28
77	57	40	59	39	21	55	33	20	48	29	16
82	71	55	67	41	19	43	27	13	42	22	12
100	78	46	55	40	24	41	31	23	34	21	10
77	58	32	52	45	32	48	33	21	82	49	21
72	51	27	59	49	34	48	33	19	94	55	22
77	56	35	82	56	32	66	39	17	60	36	16
77	58	27	88	69	43	46	27	13	56	26	12
88	69	49	87	53	24	47	28	13	36	29	21
87	69	46	55	34	23	50	32	14	73	45	30
87	54	28	51	37	24	48	32	18	88	61	23
67	51	33	54	40	26	48	36	31	63	32	14
88	62	35	54	37	23	58	33	15	59	27	10
81	48	21	88	68	36	94	49	26	49	24	13
69	48	32	88	71	43	94	87	77	83	32	13
69	44	25	88	65	40	94	78	56	61	43	20
54	39	30				94	58	28	88	46	21
61	46	28				68	44	23	56	30	15
67	53	43				58	37	21			

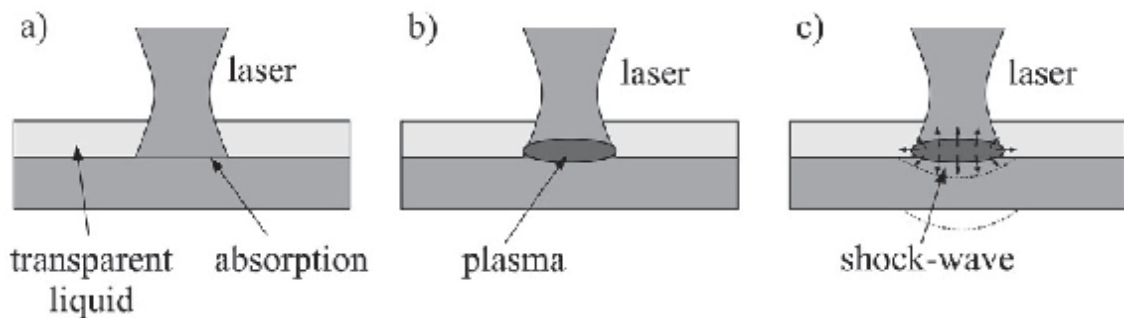


Figure (1) Shock Wave Process

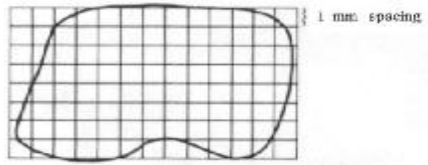


Figure (2) Shock Wave Process

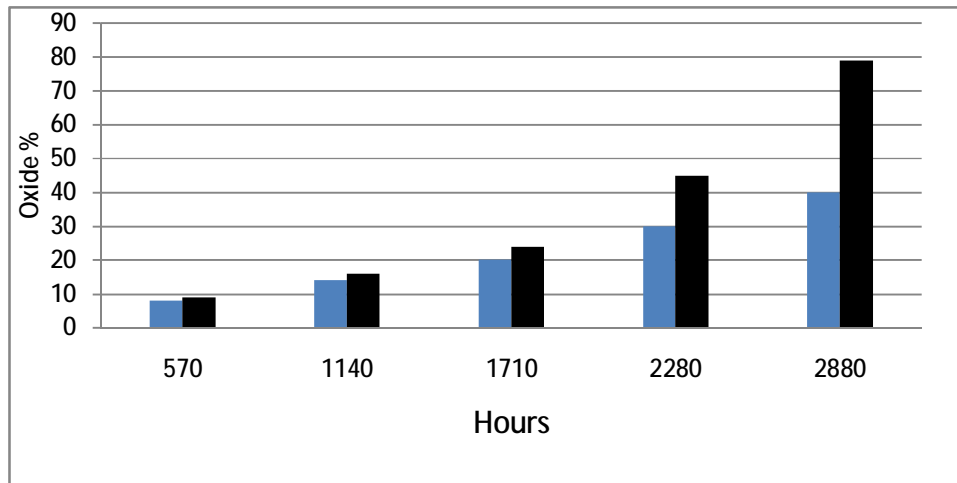


Figure (3) Percent of Oxide formation to the total surface with hour for 1 pulse

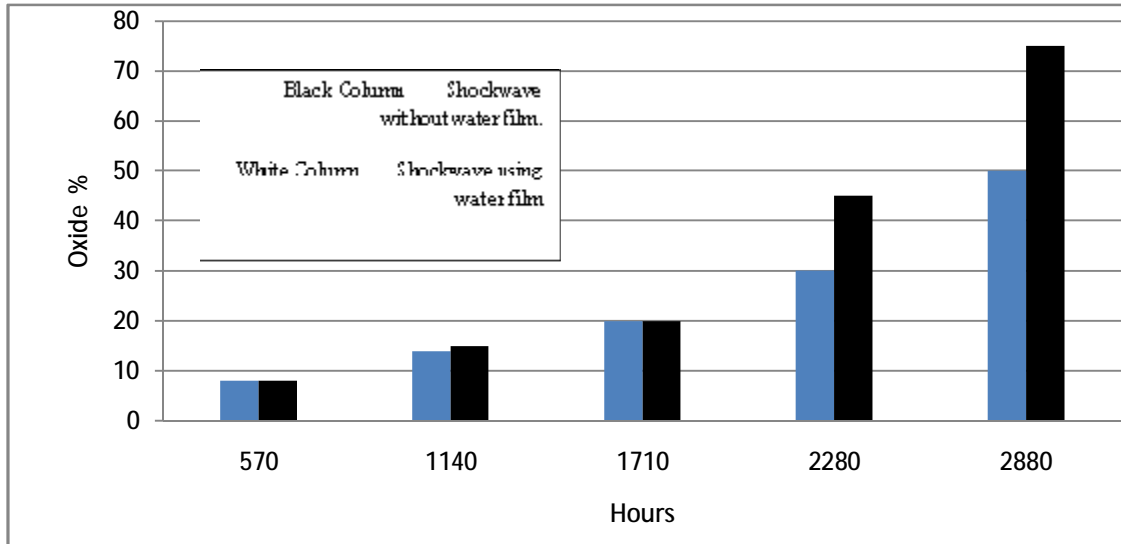


Figure (4) Percent of Oxide Formation to the total surface with Hour for 1 pulse

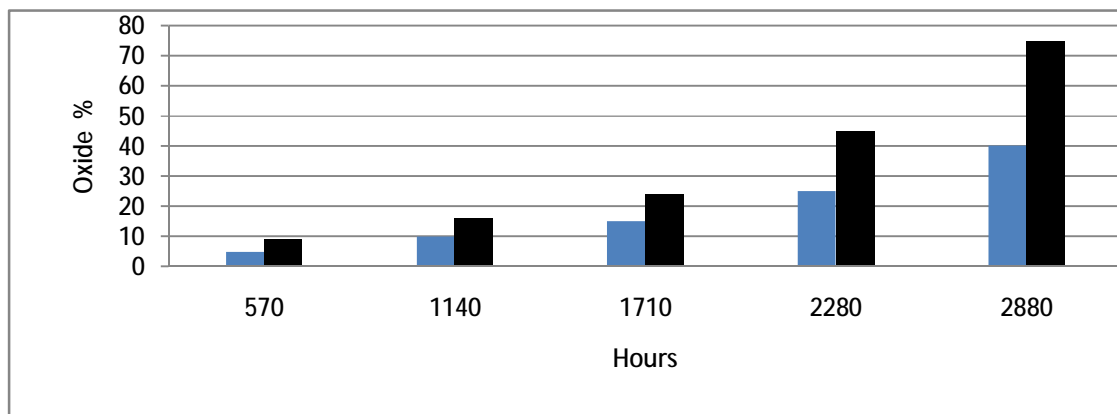


Figure (5) Percent of Oxide Formation to the total surface with Hours.2 pulses

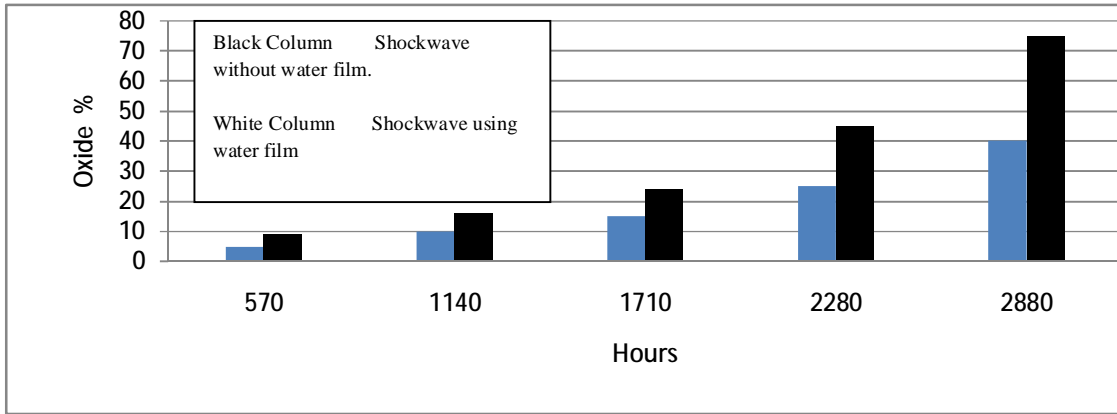


Figure (6) Percent of Oxide Formation to total surface with hours for 4 pulses

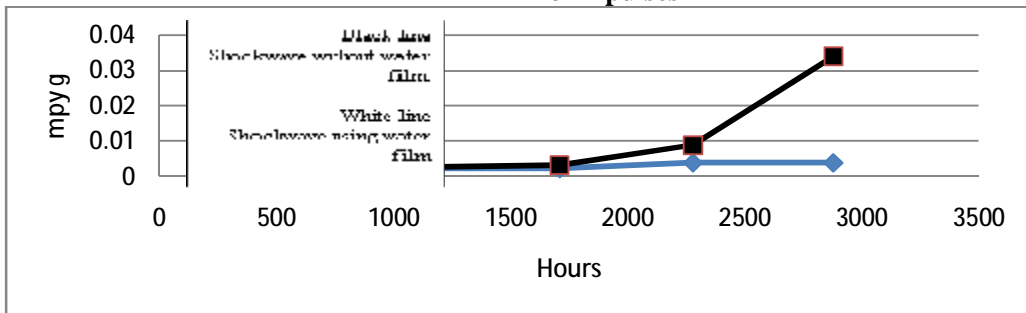


Figure (7) Percent of Oxide Formation to the total surface with Hours for 4 pulses

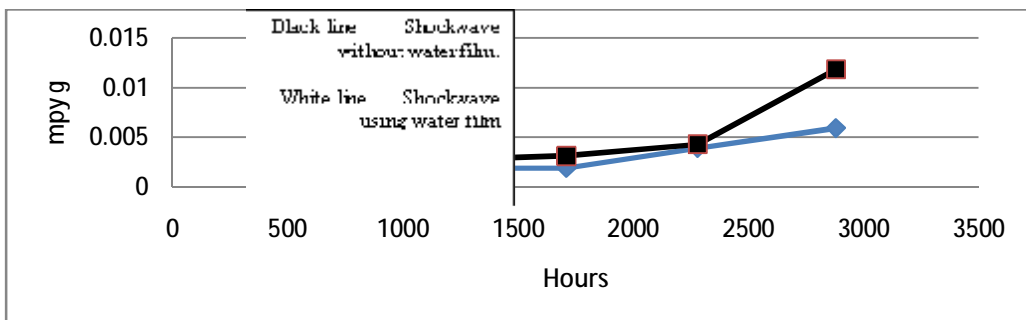


Figure (8) one pulse

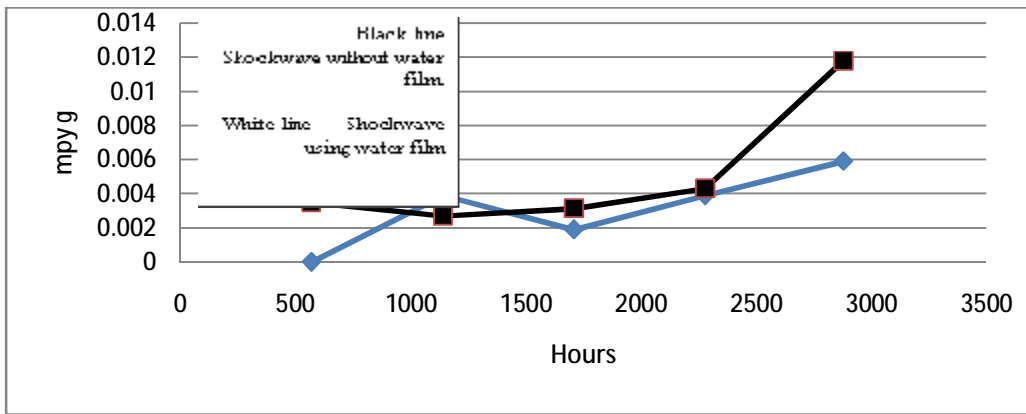


Figure (9) two pulses

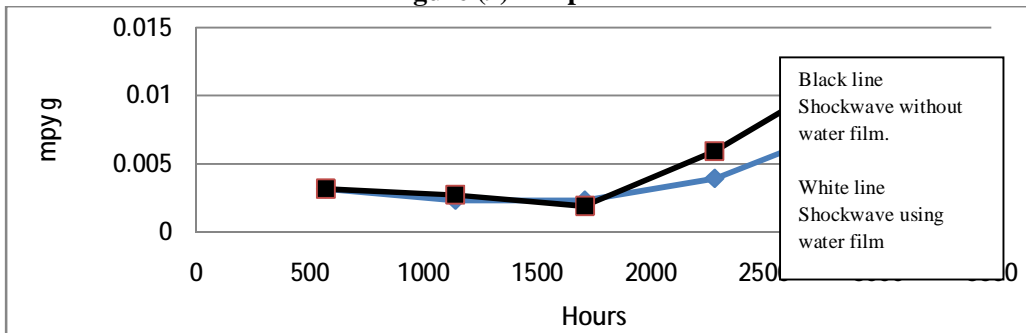


Figure (10) three pulses

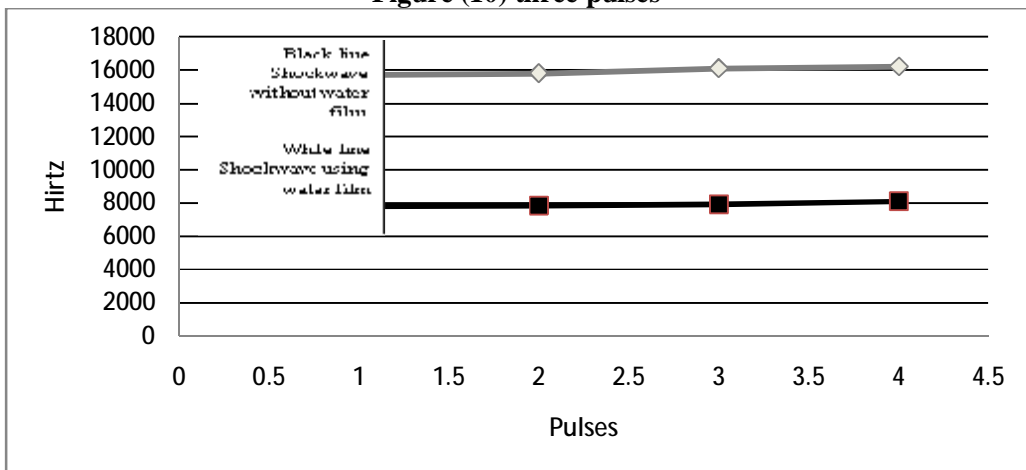


Figure (11) four pulses

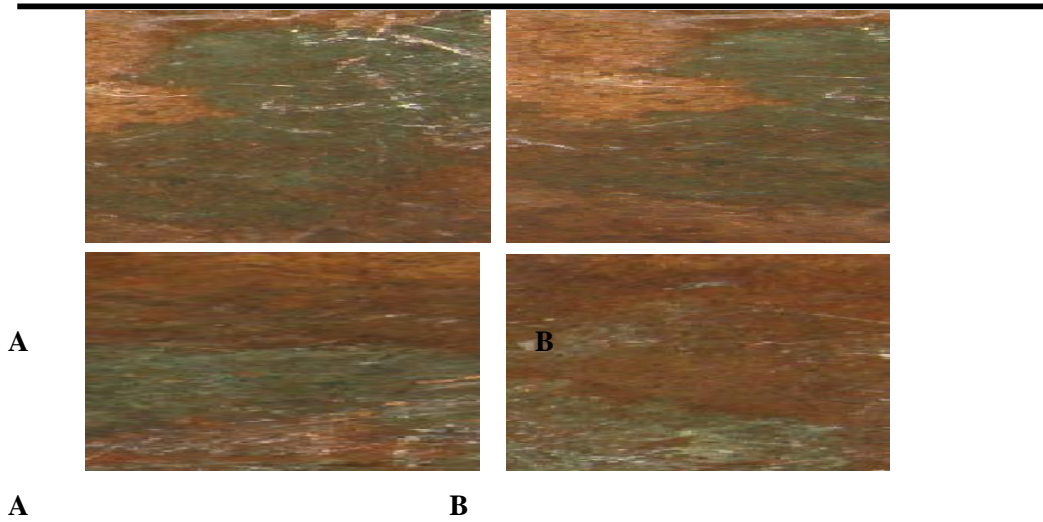


Figure (12), acoustic wave level generating from shock wave

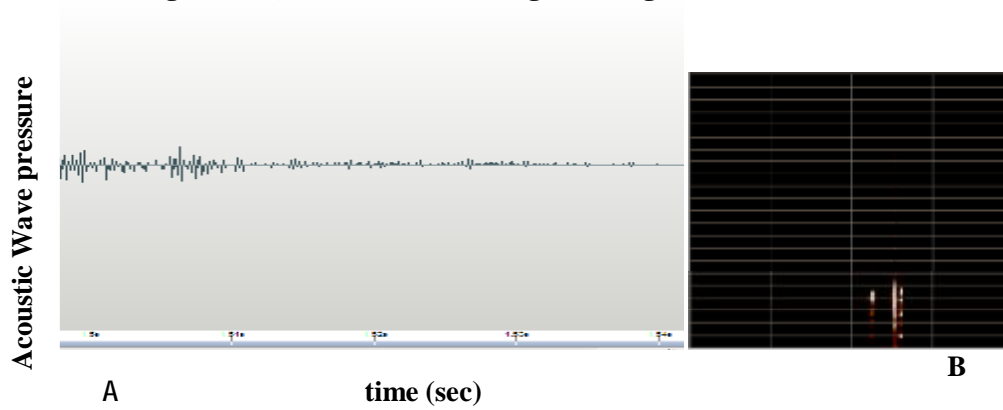


Figure (13) A the acoustic wave pressure generated from lasershock wave without water film one pulse, B is hertz level (8000) hertz,

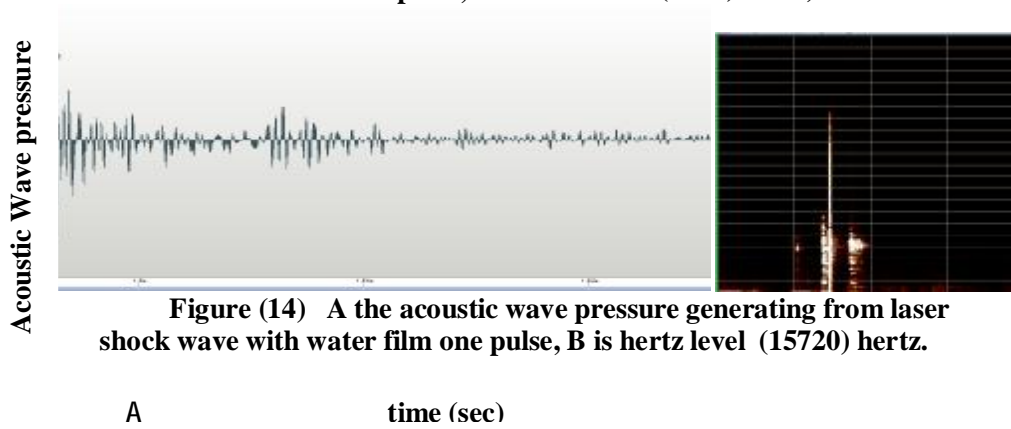


Figure (14) A the acoustic wave pressure generating from laser shock wave with water film one pulse, B is hertz level (15720) hertz.