A review on coumarin backbone: An attractive scaffold for promising bioactive compounds

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ABSTRACT

Background: Coumarin moiety is found in many naturally occurring products that have been used for many decades in traditional medicine around the world. Coumarin has distinctive physicochemical properties and can be easily transformed into a wide range of functionalized coumarins. As a result, a large number of coumarin derivatives have been designed, synthesized, and evaluated to attack a variety of pharmacological targets selectively. These targets may include various selective enzyme inhibitors and the targets, tagged as multitarget-directed ligands, found in diseases like Parkinson's and Alzheimer's, which are considered as multifactorial diseases.

Objectives: The most widely used synthetic methods leading to coumarins, besides the major biological routes for their metabolic transformations and biosynthesis, are highlighted and reviewed. Also, the focus was concentrated on some pharmacological activities of coumarin derivatives involving those related to the selective inhibition of cholinesterase and monoamine oxidase enzymes and targeting specific ligands of neurodegenerative diseases.

Conclusion: The impacts of substituents pattern/type on the selectivity and potency of the studied coumarins were explained to determine the main structural and molecular factors that may affect the activity and performance of the directed targets.

Keywords: Coumarin, Multitarget-directed ligands, Monoamine oxidase inhibitors, Cholinesterase inhibitors.

المعلومات الاساسية: يوجد جزء الكومارين في العديد من المواد التي تنتج بشكل طبيعي والتي تم استخدامها لعقود عديدة في الطب التقليدي في جميع أنحاء العالم. يتمتع الكومارين بخصائص فيزيائية وكيميائية مميزة ويمكن تحويله بسهولة إلى مجموعة واسعة من الكومارين الفعالة وظيفياً. نتيجة لذلك ، تم تصميم عدد كبير من مشتقات الكومارين وتصنيعها وتقييمها لمهاجمة مجموعة متنوعة من الأهداف الدوائية بشكل انتقائي. قد تشمل هذه الأهداف العديد من مثلوات الإنزيمات الانتقائية والأهداف ، التي تم تصنيفها على أنها روابط موجهة متعددة الاهداف، موجودة في أمراض مثل باركنسون والزهايمر ، والتي تعتبر من الأمر اض متعددة العوامل.

الغاية من الدراسة: يتم تسليط الضوء على الطرق الاصطناعية الأكثر استخدامًا المنتجة للكومارين ، إلى جانب الطرق البيولوجية الرئيسية لتحولاتها الأيضية وتخليقها الحيوي. تم التركيز على بعض الأنشطة الدوائية لمشتقات الكومارين التي تتضمن تلك المتعلقة بالتثبيط الانتقائي لأنزيمات الكولين إستريز وإنزيمات أوكسيديز أحادي الأمين ، واستهداف روابط محددة للأمراض التنكسية العصبية.

الاستنتاج: تم شرح تأثيرات نمط / نوع البدائل على انتقائية وفعالية الكومارين المدروسة لتحديد العوامل الهيكلية والجزيئية الرئيسية التي تؤثر على النشاط والأداء في الأهداف الموجهة.

الكلمات المفتّاحية: الكومارين، الاهداف المتعددة الاتجاه، متبطات انزيم تفكك الكولين، متبطات انزيم أكسدة أحادي الأمين.

INTRODUCTION

ogel was discovered coumarin **S1** from the seeds of the Dipteryx odorata tree, which has called tonka beans, in 1820, for the first time. It is also known as Coumarou, a French term ¹. Since then, many coumarins derived from plants, bacteria, fungi have been isolated, structurally characterized, synthetically modified, and evaluated for their various biological activities ^{2, 3}.

S1, whose structure is shown in Figure 1, is an oxa-heterocycle with a 2H-chromen-2one (2H-1-benzopyran-2-one or 1,2benzopyrone) that has been extensively investigated due to the presence of its skeleton in many physiologically active products and compounds ⁴.



Figure 1: Chemical backbone of coumarin.

Coumarin and some of its derivatives have been developed into medications, as displayed in Figure 2, including the anticoagulants acenocoumarin **S2**, phenprocoumon **S3**, and warfarin **S4**, which all act as vitamin K antagonists. Also, the antibiotic novobiocin S5 that is a potent bacterial DNA gyrase inhibitor, and the choleretics hymecromone (4-methyl umbelliferone) S6, and armillarisin A S7⁵.



Figure 2: The chemical structures of some coumarin-based medications.

Biosynthesis of Coumarins

Coumarins are generated naturally via the main biosynthetic route, which leads to phenylpropanoids. The enzyme phenylalanine ammonia-lyase (PAL) converts phenylalanine (from the shikimate pathway) into trans-cinnamic acid, which is then converted into the core metabolite 4'coumaroyl-S-CoA. The latter is then subsequently converted into a variety of 6 phenylpropanoids Coumarin is biosynthesized from the core metabolite via *ortho*-hydroxylation, followed by *trans-cis* isomerization of the side chain, and finally cyclization process. The *ortho*-hydroxylation, mediated via the enzyme Feruloyl-CoA 6'-hydroxylase 1 (F6'H1), is the first and most important step in this natural synthesis. Kai *et al.* (2008) suggested the radical mechanism for the biosynthesis of coumarins ⁷, which is displayed in Scheme 1.



7-hydroxycoumarin

Scheme 1: Biosynthesis of 7-hydroxycoumarin from the core metabolite 4'-coumaroyl-S-CoA as recorded by Kai *et al*.

Metabolism of Coumarins

There are two principal routes for metabolizing coumarin-based compounds, as displayed in Scheme 2, including 7hydroxylation and lactone ring-opening with oxidative decarboxylation. The latter occurs on the intermediate produced in the metabolic pathway's first step, which is coumarin 3,4-epoxide. Other probable coumarin metabolites, which are formed to a considerably lesser degree than at position 7, include hydroxylation of coumarins at positions 8, 6, 5, and 4, and 3,4-dihydrocoumarin (DHC) ⁸.



Scheme 2: Summarization of the principal coumarin metabolic pathways.

Some cytochrome P-450 isozymes (CYPs) can play a critical role in the metabolic

transformation of coumarin-based compounds. The primary families of

cytochrome P-450 that are important for medication's oxidative most biotransformation are CYP1, CYP2, and CYP3 ⁹. In human liver microsomes, the main enzyme responsible for the metabolism of coumarin-based compounds to their corresponding 7-hydroxycoumarin ¹⁰. Reduced metabolites is CYP2A6 CYP2A6 activity, which may be due to the CYP2A6's polymorphism and/or high gene multiplicity, may favor alternate coumarin metabolic routes, like the route that leads to 3-hydroxycoumarin via CYP3A4 catalysis. It was proposed that a higher amount of 3hydroxycoumarin mav stimulate the production 2-hydroxyphenyl of acetaldehyde (2-HPA), which is a cytotoxic product that might be involved in the coumarin-related toxicity ¹⁰.

Even though coumarin itself can be represented as an effective pharmacological therapy for lymphedema, several medical protocols have prohibited the utilization of this inexpensive and effective medication. The probable hepatotoxic consequences, as previously stated, that demonstrated in mice, rats, and even less frequently in humans may result in this precaution ¹¹. However, considering the prior results, patients with lymphedema who have a lower activity of CYP2A6 isozyme can be recognized and avoided coumarin treatment. The reason for that is the high possibility for metabolizing coumarin via the cytotoxic route, resulting in o-HPA¹¹. The early phenotyping of these individuals can promote a safer application of this effective and inexpensive compound, coumarin, to all other lymphedema-suffering patients with normal CYP2A6 enzyme function¹¹.

Toxicity of Coumarins

The exposure of humans to various natural coumarin-based products is high since they are abundant in fruits, vegetables, nuts,

seeds, tea, and coffee ¹². Metabolism-, carcinogenicity-, toxicity-, and safetyrelated studies concerning the coumarin itself found in foods and perfumes for cosmetic application have been reviewed. The authors of this review found that exposure from coumarin foods and cosmetics items has no risk on human health ¹³. On the other hand, other publications showed that coumarin and certain coumarin derivatives are significantly hazardous. Indeed, in the hepatocytes from various species, which include humans, hepatotoxic effects have been observed 14-16. Another important article stated that coumarin's cytotoxic effects depend on species and metabolism, so rat models can't be utilized to estimate coumarin toxicity in humans ¹⁷. Recent human investigations have found that 0.1 mg/kg body weight is the tolerated dosage intake for coumarin. So, to avoid hazardous consequences, this dose should not be exceeded ¹⁸.

Strategies for Coumarin Synthesis

The coumarin skeleton is future in a wide range of physiologically active natural medications. agrochemicals, entities. optoelectronic, and polymeric materials ^{19, 20}. As a result, enormous and ongoing efforts were made to develop novel synthetic routes and protocols, which make the crucial cyclization process to the heterocyclic ring and subsequent regioselective derivatization easier to perform. To create properly designed coumarin derivatives, a lot of effort has gone into developing more efficient and environmentally friendly synthetic methods ²¹. In recent years, the increased usage of enabling and new technologies, like novel catalysts, ultrasounds, microwaves, greener solvents, and solvent-free reactions, has made access to coumarin derivatives considerably easier ^{21, 22}. The most common and classical methods for obtaining coumarin derivatives:

the Pechmann ²³ and Knoevenagel ²⁴ reactions, which are the subjects of several coumarin-synthetic studies, concentrated primarily to improve yields, develop simple work-up procedures, and use green/recyclable catalysts and solvents. Only selected synthetic techniques that were released recently are being discussed briefly here.

Pechmann Reaction for the synthesis of coumarins

One of the most investigated synthetic methods for the synthesis of coumarin and its derived compounds is the Pechmann condensation reaction. The classical version of this coupling reaction was promoted by strong inorganic acids like HCl and H₂SO₄. The recent advances concerning this reaction type investigated various catalysts to facilitate and initiate this condensation ²³. The incoming are some examples:

A- FeCl₃ catalyzed the synthesis of coumarins

The condensing of various invigorated phenols and β -ketoesters was proceeded, utilizing 10% mol FeCl₃.6H₂O as an initiator, produced moderate-to-good yields ²⁵.

B- Tin tetrachloride-grafted on silica gel catalyzed the synthesis of coumarins

Under a solvent-free environment heated to 120° C, the titled heterogeneous catalyst (SnCl₄-grafted on silica gel) can stimulate the synthesis of substituted coumarins in good yields ²⁶.

C- γ -Fe₂O₃@HAp-Ag nanoparticles catalyzed the synthesis of coumarins

The Pechmann reaction was effectively catalyzed by an easily produced promoter, which was silver functionalized on hydroxyapatite-core-shell magnetic gammaFe₂O₃ nanoparticles. The reusable magnetic promoter forms the required coumarin derivatives, under environmentally friendly experimental settings, in high yields, and with a simple work-up procedure 27 .

D- Sulfonated carbon@titania composite loaded with Lewis acid catalyzed the synthesis of coumarins

At 60°C, carbon@titania composite loaded with Lewis acid was used as an efficient initiator in the Pechmann reaction. It gives coumarin derivatives in high yields under a solvent-free environment ²⁸.

E- Molybdate sulfuric acid (MSA) catalyzed the synthesis of coumarins

At 80°C, the Pechmann reaction proceeded in the water–dioxane blend employing molybdate sulfuric acid as a novel and effective promoter, producing the required coumarins in good yields ²⁹.

F- Ionic-liquid catalyzed the synthesis of coumarins

At 70°C and in a solvent-free environment, 1,3-disulfonic acid imidazolium-hydrogen sulfate (DSIMHS) was found to be an efficient and recyclable ionic-liquid promoter. High yields were obtained in a short period (less than 30 min)³⁰.

G- Sawdust–sulfonic acid catalyzed the synthesis of coumarins

The Pechmann reaction proceeded at 110°C and was initiated by a reusable solid catalyst named sawdust-sulfonic acid can form coumarins in high yields. This effort offered several benefits, including a simple work-up procedure, without solvent, and a shorter interaction period (less than 60 min)³¹.

Knoevenagel Reaction for the synthesis of coumarins

The Knoevenagel reaction is among the frequently examined most synthetic techniques for coumarin and its derivatives. An organic base like tertiary amine aided the conventional version of this coupling process that involves the interaction between benzaldehyde functionalized at position-2 with hydroxyl moiety and invigorated methylene. Recent research into this reaction type has looked at a variety of catalysts to promote and trigger this condensation phenotype. Here are a few samples from the newcomers ³²:

A- MgFe₂O₄ nanopromoter catalyzed the synthesis of 3-functionalized coumarins

Under ultrasound irradiation and in a solvent-free environment, $MgFe_2O_4$ is an

effective nanopromoter that can initiate the condensation reaction between numerous 1,3-dicarbonyl and salicylaldehydes compounds utilizing Knoevenagel reaction. This strategy has brief reaction times, a simple work-up procedure, and high yields ³³.

B- Potassium phthalimide catalyzed the synthesis of 3-functionalized coumarins

An efficient, fast, and environmentally friendly procedure for high-yielding, potassium phthalimide can promote the synthesis of 3-carboxy- and 3-cyanocoumarin phenotypes S8, as displayed in Salicylaldehyde-based Scheme 3. derivatives were reacted with invigorated methylene compounds in an aqueous media and at room temperature for several hours using the aforementioned catalyst and under mild reaction settings ³⁴.



Scheme 3: Synthesis of 3-functionalized coumarins using potassium phthalimide as a catalyst.

Pharmacological Activities of Coumarins

Coumarins have impressive pharmacological effects depending on their basic backbone (for example, simple coumarin, bis-coumarin, fused polycyclic coumarin) and substitution pattern ³⁵. The pharmacological actions that have received the most attention including antiviral ³⁶, 37, 38 antifungal antibacterial anti-39 anticoagulant 40 inflammatory 41 42 anticancer antithrombotic antimutagenic ⁴³, antioxidant ⁴⁴, cytotoxic ⁴⁵, CNS stimulant ⁴⁶, cholinesterase (ChE) ⁴⁷, 46 monoamine-oxidase (MAO) 49 lipoxygenase ⁴⁸, and cyclooxygenase

activities. The coumarin ring has structural and physicochemical characteristics that making it easy for binding to many target sites.

Because the coumarin ring is aromatic, lipophilic, and planar, it can combine with physiological counterparts, primarily the lipophilic binding sites in proteins, by forming π - π stacking (non-covalent formed-forces generated among aromatic rings) and hydrophobic interactions with aromatic amino acids such as phenylalanine, tryptophan, and tyrosine. Also, coumarins may attach to positively-charged amino acids via strong ion-dipole interaction ⁵⁰.

Besides, the coumarin's lactone group can offer the capacity to form strong-polar connections, such as dipole-dipole interactions and hydrogen bonds.

In some cases, coumarins can acylate the protein targets, as suggested for a particular enzyme's covalent inhibition mechanism. The lactone ring can also be opened by esterase-enzyme phenotypes, and biological activity may be caused by molecules that arise from that hydrolysis. In this situation, coumarins are bio-activated to release the actual active metabolites, acting as prodrugs 50. Several natural coumarins have been reported to have this mode of action for inhibiting anhvdrase. carbonic which possesses the catalytic activity of esterase ⁵¹. Pharmacological activities of coumarins on selected targets are discussed below.

Coumarins as ChE Inhibitors

Acetyl- and butyryl-cholinesterase (ACh-E and BCh-E), mainly ACh-E, catalyze the breakdown of acetylcholine (ACh) neurotransmitters at the synaptic gap of cholinergic neurons. Their inhibition can restore ACh levels in the nervous system, so ACh-E inhibitors were used in the treatment of Alzheimer's disease (AD), which led to rising interest in the development of new

ACh-E/BCh-E inhibitors ⁵². Coumarin derivatives, both synthetic and natural, are a well-studied group of compounds that function as ACh-E/BCh-E inhibitors, with several reviews published on the subject recently ⁴⁷. To achieve the maximum inhibition of these enzyme phenotypes, a highly trendy step in the designing of ChE inhibitors was used, which led to conjugated compounds having the moiety of coumarin coupled to a well-known ChEs inhibitor, usually tacrine or donepezil. Although there is no complex for co-crystallization of human ChEs (hChEs) with these kinds of coumarin hybrids yet, inhibition kinetics and docking studies propose that such coumarinbased molecules having a binding posture, typically extending from the peripheral (PAS) to the catalytic active site (CAS) of the enzyme, the term "dual binding site (DBS) inhibitor" was coined to describe these compounds ⁵³. Also, the crystal structure of the combination of Torpedo californica A-ChE with aflatoxin confirmed that the coumarin ring preferentially binds to a peripheral active site (PAS) of the tested enzyme ⁵⁴. As shown in Table 1, many coumarin-based compounds were investigated and identified ChE as inhibitors.

Symbol	Compound	Activity	Characteristics	Structure
S9	3,4-Dimethyl-7- hydroxycoumarin attached to an edrophonium-like moiety (coumarin–edrophonium heterodimers) ^{55, 56} .	DBS bovine (bAChE) inhibitors	Ammonium salts with low nM IC ₅₀ values. Dipole-dipole and ion-dipole interactions have been proposed for coumarin- edrophonium heterodimers, high bAChE over esBChE selectivity.	HO O O O O

S10	6,7-Dimethoxy-3- substituted coumarin derivative linked to 3- hydroxy-N, N- dimethylanilino via a suitable linker ⁵⁷ .	bAChE- selective inhibitors	The IC ₅₀ was 0.236 nM, and the selectivity of AChE/BChE was high (SI > 300,000).	HO CH2)4 O O
S11-15	Coumarin–pyridinium derivatives ⁵⁸ .	electric eel- (eeAChE) inhibitors	Quaternary benzylammonium salt, IC_{50} s in nM-pM range, introducing substituent in position 3 for effective binding with the peripheral active site.	R ^{II} R ^{II} R=H, 6-Br, 8-OMe R`=H, Me, OMe, Hal
S16	Donepezil-like coumarin, 6,7-dimethoxycoumarin derivatives containing a protonatable benzylamino group, linked to position 3 via various linkers ⁵⁹ .	bAChE- selective inhibitor	IC ₅₀ = 7.6 nM, a mixed-type inhibitory activity was found, proving the binding with both the PAS and CAS of bAChE.	
S17	AP2238, class of 3- benzylaminocoumarins, bearing the 6,7- dimethoxy group ⁶⁰ .	hAChE- selective inhibitor	One of the donepezil-like compounds, which was early released and advanced to pharmacological and biochemical testing.	
S18 and S19	3-Carbox amidocoumarins with small substituents in coumarin ring at positions 8, 7, or 6 ⁶¹ .	eeAChE- selective inhibitors	Donepezil–coumarin hybrids, high selectivity of AChE/BChE, with nM binding affinity on eeAChE, also showing promise neuroprotection effect.	$R \xrightarrow{H}_{U} O O O O O O O O O O O O O O O O O O O$
S20-22	Tacrine–coumarin hybrids ⁶² .	hChEs- inhibitors	Potent ChE inhibitors designed by using a recurrent approachs.	HO O O N HO N HO N HO N HO N HO N HO N

Coumarins as MAO Inhibitors

Endogenous and exogenous amines, including neurotransmitters, are oxidatively deaminated by monoamine oxidase enzymes (MAOs). In humans, there are two types of MAO enzymes: MAO-A and MAO-B, which differ in substrate selectivity and inhibitory sensitivity ⁶³. Selective MAO-B inhibitors are used nowadays in conjunction

with levodopa to treat Parkinson's disease, whereas MAO-A inhibitors are used to treat depression ⁶⁴. Knoll AG and BASF published a joint paper in 1994 that reported for, the first time, the inhibiting action of a previous series patented, including 7-aryl sulfonyloxycoumarins and 7-arvl alkoxycoumarins as MAOs inhibitors with high selectivity ⁶⁵. Such exceptional selectivity and activity were examined

further by many researchers, as shown in Table 2.

The easily functionalize coumarin backbone synthetically enables the investigation of various substitution patterns. The resolution of co-crystallized complexes of human MAOs with many reversible/irreversible inhibitors by X-ray crystallography led to an advance in the design of MAOs inhibitors ^{66,} ⁶⁷. By using the structural information obtained from the studies of X-yay as a a target-dependent design guide, of expanded sequences of selective inhibitors of MAO-B was created, by retaining the substituent 7-m-chlorobenzyloxy, to ensure selectively and desirable binding to MAO-B and trying to introduce charged or polar substituent at position 4. B/A selectivity and MAO-B affinity were preserved in nearly all of the proposed compounds, which also demonstrated high solubility in water and low lipophilicity, all of which are essential requirements for progressing to pre-clinical and clinical investigations ⁶⁷.

Also, to discover and justify the most significant molecular factors for high selectivity of MAO A/B and MAO affinity, as well as to give drug-like characteristics to recently developed inhibitors, different computational studies were carried out (computer-promoted molecular design) ⁶⁸. Lately, molecular dynamics (MD) simulations used to check the interaction of hMAO-B and -A with selective coumarinbased inhibitors at the molecular level. A significant finding has been the discovery of water-mediated H-bond between Flavin adenine dinucleotide (FAD) cofactor and the selective MAO-B 7-benzyl oxycoumarin, an interaction that was not anticipated by docking-studies of similar analogs $^{\overline{69}}$.

Table 2: Coumarin derivatives act as MAO inhibitors

Symbol	Compound	Activity	Characteristics	Structure
S23 and S24	7-Benzyloxy and 7- arylsulfonyl oxycoumarins ⁷⁰ .	MAO- inhibitors	IC ₅₀ s in the nM range, selective and potent.	X II Y=CH2, MAO B selective Y=SO2, MAO A selective
S25 and S26	Ester derivatives of 7- hydroxy-3,4- dimethylcoumarin ⁷¹ .	rMAO- inhibitors	The <i>ex vivo</i> studies on this group of coumarins were carried out on the rat's brain and liver for the first time. The half-life of esters hydrolysis in a buffer correlated with the R group's steric hindrance.	
S27-34	Series of natural and newly synthesized geiparvarins, which are 7- substituted coumarins ⁷² .	rMAO- inhibitors	Strong selectivity of MAO B/A and high rat MAO-B inhibitory action in low- to the sub-µM range.	R_1 R_1 R_2 R_1 R_2 R_1 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_2 R_2 R_2 R_2 R_2 R_2 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_3 R_4 R_4 R_3 R_4 R_3 R_4 R_4 R_3 R_4 R_4 R_3 R_4 R_4 R_3 R_4 R_4 R_3 R_4 R_4 R_4 R_3 R_4 R_4 R_5 R_4 R_4 R_5 R_4 R_4 R_4 R_4 R_4 R_4 R_4 R_4 R_5 R_4 R_4 R_4 R_4 R_4 R_5 R_4 R_4 R_5 R_4 R_5 R_4 R_5

S35 and	3-Carbox amidocoumarins	hMAO-	Different selectivity of MAO	0
S36	73	inhibitors	B/A, high inhibitory potency of	A A Å BA
			human MAO, with IC50s	N ¹¹²
			ranging from μM to sub- μM .	
			A decrease in MAO-B affinity	H ₇ T 0 0
			occurred due to the addition of	R ₈
			the benzyloxy group in position	R1, R8= H, Me
			7.	R/=n, bno R2= (cyclo)alkyl, (hetero)aryl
S37-46	3-Acylcoumarin derivatives	hMAO-	Inhibitors with low potencies,	0
	74.	inhibitors	except 3-carboxy	Baaal
			hydrazidocoumarin and 7-	e
			benzyloxy-3-ethylester	
			derivatives, having good	$R_7 \rightarrow 0^{-1} 0$
			hydrolytic stability and strong	Ř ₈
			inhibition of human MAO-B	R6, R8= H, Me, MeO,OH, Hal
			$(IC_{50} \text{ equal to } 3.2 \text{ nM}).$	R7= H, OH, (cyclo)alkoxy, BnO, NEt2
				X= Me, Ph, OH, OEt, NHNH2, NHNHAr
S47-54	3-Carbamylcoumarins ⁷⁵ .	MAO-	Higher selectivity of MAO B/A	H e
		inhibitors	and lower inhibitory potencies	^N √ ^O B
			in comparison with /-benzyl	
			oxycountarins analogs.	
				R= (balo)Alk Ph Bn
				re- (naio)/-ux, rn, bri
S55-64	3-Phenyl-6-substituted	hMAO-	Potent and selective human	Ba
	coumarins ⁷⁶ .	inhibitors	MAO-B inhibitors and IC50s in	Bu J Bu
			the sub- μ M to the sub-nM	112
			range. The addition of	R ₆
			coumaring seemed tolerable	H ₅
			except for the insertion of a	\sim
			bulky group.	
				R _x = H, Me, OH, MeO, Br
S65 and	3-Pyridazinyl	hMAO-	High selectivity of MAO B/A,	^ X
S66	coumarins ⁷⁷ .	inhibitors	an affinity for human MAO-B	r i i i i i i i i i i i i i i i i i i i
			in the µM range, and good	
			predicted pharmacokinetic	BU Y Y N
			properties (ADMET).	
				÷ 0 °0
				R= Me, MeO, (6, 7, or 8)
				X= Br, CI, MeO
S67	7-[(3-Chlorobenzyl) oxy]-4-	rMAO-	High B/A selectivity and rat	Н
	[(methylamino) methyl]-	inhibitor	MAO-B inhibitory potency.	~ ^N ~
	2H-chromen-2-one (NW-		Good pharmacokinetic	
	1772 compound) ^{10,79} .		and high blood brain barrier	
			(BBB) permeation, IC50 of	
			MAO-A= 5.94μ M and MAO-	
			B=13nM.	L l
				~

S68-70	New 7-benzyloxycoumarin derivatives (computer-aided design) ⁸⁰ .	Potent hMAO-B- inhibitors	Excellent selectivity of MAO B/A and the activity on human MAO-B in the nM to the sub- nM range. These compounds developed through target- and ligand-depended screening of well-known selective inhibitors of MAO.	$\begin{array}{c} & & & \\ & & & \\ & & & \\ & & & \\ & &$
S71	Coumarin derivatives designed by using Computer-aided technologies ^{68, 81} .	Highly selective hMAO-A- inhibitors	Derived by using a combined QSAR-CN (complex networks) model approach and MARCH- INSIDE approach (computational method).	
S72 and S73	Coumarin derivatives designed by using Computer-aided technologies ^{68, 81} .	Highly selective hMAO-B- inhibitors	Derived by using a combined QSAR-CN (complex networks) model approach and MARCH- INSIDE approach (computational method).	Br O O O Cy= -(CH=CH)2-, -(CH2)3-

Coumarins as Multitarget-Directed Ligands

The use of a multi-target drug therapy to neurodegenerative and treat other complicated illnesses become has а "revolution" in drug development. Because of their multiple etiology, multifactorial are usually treated diseases with а medication cocktail, which has a higher risk of drug-drug interactions and toxicity. The multi-target strategy is based on the idea of one drug, several targets, which intends to provide a single-drug therapy with multiple pharmacological actions that may be used to treat the same illness ⁸².

The benefits of this monotherapy include primarily the absence of drug-drug interactions, increased compliance of patients who are taking only one therapy for

their condition, and the simplicity of ADMET screening and therapeutic and pharmacological characterization. However, such a multi-target profile necessitates a correct balance of potencies to achieve the required pharmacological activity for every target addressed. The simplest method of preparing multi-target ligands is to attach two pharmacophoric moieties, each of which responsible is for different biopharmacological actions, via a linker that can be cleaved metabolically (conjugationtype) or through a covalent bond that is formed directly (fusion-type). A higher advanced multi-target ligand design involves merging (hybridizing-type) two or more functional groups into a novel molecular unit, capable of displaying the original actions of both moieties ^{83, 84}, as shown in Figure 3.



Figure 3: Different strategies of multi-target drug design.

The disadvantage of creating conjugated or fused molecules is that it frequently leads to big molecules with low water solubility, high lipophilicity, high molecular weight, and a large number of rotatable bonds. As a result, these compounds are likely to have poor pharmacokinetic characteristics, including limited bioavailability and higher susceptibility to being substrates for detoxification systems. On the other hand, hybrid molecules may maintain more druglike properties and may be easier to create as hit drugs for pharmacological investigations ^{85, 86}.

The significant interactions of the coumarin core inside the binding sites of ChEs and MAOs enzymes lead to molecular conjugates having this skeleton, which is also known as dual ChE–MAO inhibitors ⁸⁷. Some coumarin derivatives that can act as multitarget-directed ligands are reported in Table 3.

Symbol	Compound	Activity	Characteristics	Structure
S74	7-Benzyloxycoumarins ⁸⁸ .	Dual eeAChE- rMAO-B inhibition	Act as mixed/noncompetitive electric eel AChE inhibitors in the 3–100 μM range and rat MAO-B selective.	X II Y=CH2, MAO B selective
S75-85	Protonatable 7-substituted coumarins, a flexible alkoxy chain was used to connect the basic N-benzyl group to position 7 ⁸⁹ .	Dual ChE– MAO-B inhibition	Good inhibitory activities at MAO-B, AChE, and BChE, but low selectivity.	X N n 0 Me(Et) n= 3-6, X= H, Cl, CN

Table 3: Coumarin derivatives act as multitarget-directed ligands

<u>586-90</u>	Protonatable 7-substituted coumarins, A more rigid 7- benzyloxy moiety was used to connect the basic N-benzyl group to position 7 ⁸⁹ .	Dual ChE– MAO-B inhibition	The presence of CH ₂ OH group in position 4 of the coumarin led to higher selectivity of MAO B/A, better balance of MAO-B and AChE activities, and good ADMET properties. This group of congeners has the potential to be developed into hit structures in the future. Potency in the nM range for	R_{A} R
S91 and S92	Protonatable 7-substituted coumarins, using N-benzyl piperidine substituent (Donepezil's pharmacophoric moiety) ⁹⁰ .	Dual ChE– MAO-B inhibition	hAChE. High selectivity and potency, good ADMET properties, especially high water solubility and BBB permeability, high cytoprotective effect against oxidative stress, and low cytotoxicity.	Bn-N n0 00 n=0, 1
S93 and S94	Protonatable 7-substituted coumarins. Highly close congeners of S91 and S92 compounds ⁹¹ .	Dual ChE– MAO inhibition	Excellent activity profile as S91 and S92 compounds.	Bn X 0 0 0 0 R3= H, Cl, CN R4= H, Me, CF3 X= - NH Y= CH N
<u>S95</u>	7-Substituted coumarins using a linker with second basic nitrogen ⁹² .	Dual ChE/MAO inhibitors	The multi-target action was retained, but the potency was reduced.	
<u>\$96-99</u>	7-Substituted coumarins using a charged N-benzyl pyridinium group ⁹³ .	Dual ChE/MAO inhibitors	The multi-target action was retained, but the potency was reduced. The addition of charged residues, in particular, had a negative effect on hMAO-B affinity.	R N ⁺ R= Me, F, Br, NO2
S100- 109	Coumarin–tacrine conjugates ⁹⁴ .	Multitarget- directed ligands	Good activity profiles, although protonatable piperazine is present in a spacer, the length of the linker allows for a more marked increase in the affinity of hMAO-B and hBChE.	R4 R3 R3 R3 R3 R3 R3 R3 R3 R3 R3 R3 R3 R3

S110- 129	Coumarin–tacrine conjugates (7-hydroxy coumarin–tacrine conjugates) ⁹⁴ .	Multitarget- directed ligands	The piperazine ring in a spacer connected to tacrine by an amide-bond. Efficient inhibitors of eeAChE/esBChE, with metal-chelating, and anti- amyloid properties.	$\begin{array}{c} R_4 \\ R_4 \\$
\$130- 133	Coumarin–tacrine conjugates (3-carboxy coumarin–tacrine conjugates) ⁹⁵ .	Multitarget- directed ligands	Non-selective and potent inhibitors of hAChE/BChE, with extra inhibitory action of amyloid-β aggregation and β- secretase (BACE1).	R ₆ R ₇ 0 0 N 1 N 1 N N N N N N N N N N N N N
\$134- 137	3-Propargylamine derivatives of coumarin (the basic group was shifted topologically to the coumarin ring position 3) ⁹⁶ .	Dual ChE– MAO inhibitors	Act by inhibiting MAO-B through another mechanism while retaining good dual AChE-MAO-B inhibitory action, with cytoprotective effects and high BBB permeability.	R ₇ 000 R6, R7= H, MeO

CONCLUSION

The main structural and molecular factors that affect the activity and performance in the directed targets. Such knowledge can help in designing new coumarins with more selectivity and better pharmacological action. NMR spectrophotometer and X-ray crystallography are used to determine their three-dimensional structures. The safety of coumarins is another factor to consider. The danger may occur just at extremely high coumarin doses, which is difficult to achieve in typical diets. Despite recent significant discoveries, the development of selective and potent coumarins remains a significant promising goal for pharmaceutical chemists.

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