

Available online at: http://www.basra-science-journal.ogr



ISSN -1817 -2695

Received 8-9-2016, Accepted 23-4-2017

QSAR STUDIES OF AMINO ACIDS CONJUGATED 2-AMINO-ARYLTHIAZOLE AS ANTI-BACTERIAL

MAHIR A. JALAL^a and WISAM A. RADHI

Department of Chemistry and Polymer Technology, Polymer Research Center, University of Basrah,

Basrah, Iraq.

E-mail: amahir_asaad@yahoo.com

ABSTRACT

Quantitative structure-activity relationships (QSAR) of 16 thiazole derivatives against five types of bacteria, namely: Klebsiella pneumoniae, Staphylococcus aereus, Escherichia coli, Bacillus substilis and Pseudomonas areginosa, was carried out. Physiochemical parameters were obtained computationally by optimization of chemical structures at minimum energy using molecular mechanics (MM+) theory and the semi-empirical molecular orbital (AM1) method. Then, multiple linear regression (MLR) analysis used to correlate changes in chemical structure and experimental antibacterial activity. Moreover, it is found that some descriptors are in strong dependences to provide good information and play important roles to evaluate antibacterial activity of thiazole derivatives. Results, 20 equations were found statistically with correlation coefficient rang (0.806-0.955). In addition, within the current QSAR study, the quality of the discriminant models was determined on the basis of correlation coefficient (R²), Fisher ratio (F), and standard error (S). Consequently, five good models with high correlation coefficient ($R^2>0.9$) were nominated as valuable theoretical base to improve 15 other subset predicted structures of amino acids conjugated 2-amnio-arylthiazole demonstrating anti-bacterial activity, and most of them showed good antibacterial activities against positive gram of bacteria.

Keywords: Thiazole, Amino acids, Anti-bacterial, (OSAR) Model

1. INTRODUCTION

Quantitative structure property/activity relationships (QSPR/QSAR) is a good analysis technique used to correlate chemical structure and specific chemical properties,

such melting point, boiling point, glass transition temperature TG and biological activates, by modeling between calculated descriptors obtained from molecular structures and desired property which is got

experimentally, and the results are derived models that help predict to properties/activities of untested chemicals [1]. When little or no more actual databases are available, QSPR/QSAR is scientifically expecting tool for great property/activity is for uncharted chemicals. These studies provide good interpretations and have successfully assisted chemists to reduce their working time, mistakes, and errors in designing of compounds with desired activity/property [2]. QSPR/QSAR studies generally involve two steps: first, descriptors (physicalchemical parameters) are generated which encode for chemical structural information; and second, a statistical regression method correlates changes in structure with changes in chemical properties or biological activity. The compound in the training set (i.e., the selected construct data set to QSPR/QSAR model) should be diverse in both chemical structure and property/activity to ensure a statistically robust model. The method typically assumes that chemicals function by a common mechanism. The model is then validated by predicting the property/activity for test set (i.e., a group of chemicals not included among the trainingset compounds). One validated, QSPR/QSAR models can be used to predict desired property/activity of untested chemicals [3].

In the last decades, QSAR models have been applied extensively in various areas and have taken great attention in medical chemistry such as toxicity prediction of single-celled in natural water [4], designing chemicals as narcotics and predict their toxicity [5], subjecting natural isolated polyol esters on QSAR to find optimal narcotic activities and insecticidal activities [6], and improving anti-cancer activates for 12 anti-cancer Schiff-base ligands [7]. Also, many researchers reported of SQAR studies on

steroids types. For example, several Correlation of lipophilicity degree of aminosteroids derivatives with descriptors. partition coefficient (LogP) and π , was studied [8,9] In addition, studying the effect steroid-based drug structures penetrating the bilayer membrane, and correlate with **Passive** Diffusion, P-Glycoprotein Active Efflux and P-Glycoprotein inhibitor by using QSAR method [10]. P. Bhattacharya and et al, subjected 30 thiazole and thiadizole derivatives to build potent and selective human adenosine A3 receptor antagonists using QSAR [11, 12]. Potent tri-substituted thiazole derivative was designed by other [13], for cancer therapy as serine-threonine protein kinase inhibitor. QSAR analysis of derivatives thiazole with antihistamine activity was also carried out, and then elaborated new thiazole derivatives [14]. Tube dilution method used to evaluate experimentally 12 molecules of 2,4disubstituted thiazole as antimicrobial agents and QSAR analysis showed excellent binding for new thiazole derivative [15]. Antiplatelet activities of 20 thiazole derivatives were correlated successfully using QSAR models [16].

It is thus a primary object of the present research to build OSAR models of 16 molecules set of thiazole derivatives, which amino acids conjugated 2-amnioarylthiazole, and study the relationship of obtained descriptors such as hydrogen bonding, energy gap, dipole moment, and etc. with in-vitro antibacterial activities against three types of negative bacteria: Klebsiella pneumoniae, Staphylococcus aereus, Escherichia coli, and two types of positive bacteria: Bacillus substilis and Pseudomonas areginosa. Finally, results used to predict antibacterial activities of 15 unprepared molecules of thiazole-amino

acids derivatives.

2. METHOD GEOMETAY OPTIMIZATION

Theoretical calculations were performed by PCGamess, running on a Pentium V PC-CPU 3.4GHz. The geometries of the compounds were optimized first at level (MM+) by molecular mechanics force field theory and then by (AM1) semi-empirical method [17-18]. Then, antibacterial activities of conjugated thiazole-amino acids were regressively utilized for modeling by using multiple liner regression (MLR).

3. EXPERIMANTAL

Antibacterial activities of 16 thiazole molecules against five type of bacteria have been determined experimentally by agar well diffusion method, which a circular well of diameter 6 mm was made exactly at the center of the plates by using cork borer and each well was filled with 0.1 mL of the thiazole-test solution (10mg/mL)[19, 20]. Streptomycin was tested at all agars as standard for thiazole antibacterial. The structures of thiazole derivatives are shown in Figure1, and their experimental inhibitory zones (mm) were also tabled in Tabel 1.

Table1: List of Experimental Inhibitory Zones (i.z.) of 2-amnio-arylthiazole Derivatives Against Five Types of Bacteria.

		Inhibitory zone (i.z.) diameter (mm)							
		Gra	m negative bac	teria	Gram positive bacteria				
Molecula No.	r Molecular name	Klebsiella pneumoniae	Staphylococcus aereus	Escherichia coli	Bacillus l substilis	Pseudomonas areginosa			
1	4-phenylthiazol-2-amine	1	1	2	1	1			
2	4-(4-chlorophenyl)thiazol-2-amine	1	1	3	2	2			
3	N-(4-phenylthiazol-2-yl)pyrrolidine-2-carboxamide	3	2	4	2	2			
4	<i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)pyrrolidine-2-carboxamide	3	2	4	4	5			
5	2-amino-3-phenyl-N-(4-phenylthiazol-2-yl)propanamide	3	2	5	2	3			
6	2-amino- <i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)-3-phenylpropanamide	4	3	6	6	4			
7	2-amino-3-(4-((2,6-dichlorobenzyl)oxy)phenyl)- <i>N</i> -(4-phenylthiazol-2-yl)propanamide	4	3	5	2	3			
8	2-amino- <i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)-3-(4-((2,6-dichlorobenzyl)oxy)phenyl)propanamide	8	4	7	7	5			
9	2-amino-3-(1 <i>H</i> -indol-3-yl)- <i>N</i> -(4-phenylthiazol-2-yl)propanamide	6	4	6	3	5			
10	2-amino- <i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)-3-(1 <i>H</i> -indol-3-yl)propanamide	- 9	6	8	8	8			
11	2-amino-3-(1-((benzyloxy)methyl)-1 <i>H</i> -imidazol-4-yl)- <i>N</i> -(4-phenylthiazol-2-yl)propanamide	3	2	3	2	4			
12	2-amino-3-(1-((benzyloxy)methyl)-1 <i>H</i> -imidazol-4-yl)- <i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)propanamide	3	4	4	4	4			
13	2-amino-3-(benzyloxy)- <i>N</i> -(4-phenylthiazol-2-yl)propanamide	2	2	3	2	2			
14	2-amino-3-(benzyloxy)- <i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)propanamide	4	2	4	4	3			
15	2-amino-5-(3-nitroguanidino)- <i>N</i> -(4-phenylthiazol-2-yl)pentanamide	3	2	4	2	3			
16	2-amino- <i>N</i> -(4-(4-chlorophenyl)thiazol-2-yl)-5-(3-	6	4	5	5	5			

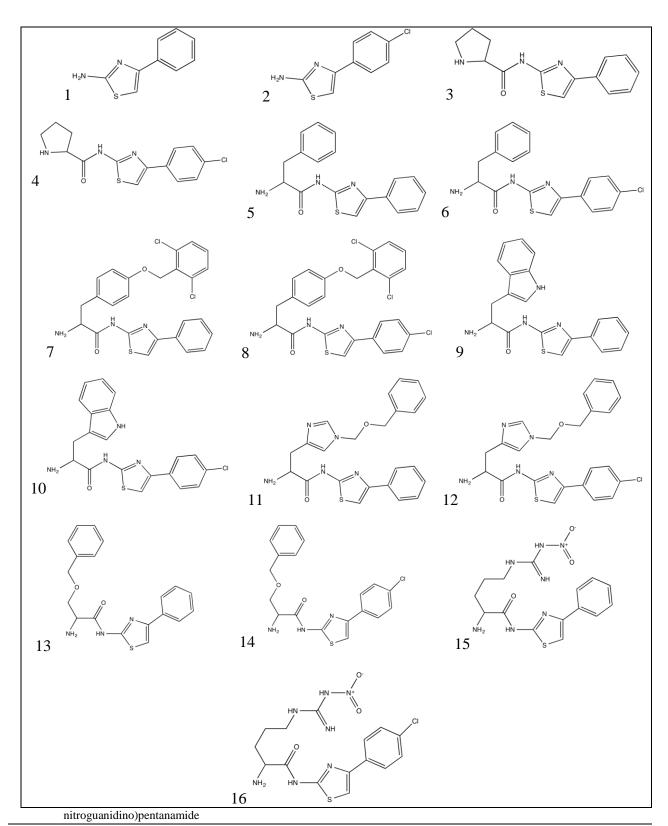


Figure 1: Chemical Structures of 16 Thiazole Derivatives.

4. RESUILTS AND DISCUSSION

The derivation of theoretical molecular descriptors proceeds from the chemical structure of the compounds. The training sets of prepared molecules are summarized in Table 2. Four QSAR models for each bacteria type were produced, overall, 20 models for all bacteria types are built up

using a training set of 16 molecules and parametrically depicted in Eqs. 1-20 with correlation coefficient (R²=0.801-0.955). It is well known that a high predictive ability can be obtained when MLR model is one that has high R² and Fisher ratio (F) values, low standard error (S) and least number of descriptors [21,22].

Table 2: Calculated Physio-Chemical Descriptors of the Thiazole Compounds

Molecule No.	H.E.	Pol.	Mass	Ref.	Log p	Vol.	S.A.	S.G.	НОМО	LUMO	E. gap	N.E.	T.E.	D.M
1	-9.75	20.26	176.24	56.87	0.35	538.89	270.42	353.01	-8.2788	-0.1854	-8.0934	276.240	-68.709	2.183
2	-9.4	22.19	210.68	61.59	0.13	582.31	305.49	367.92	-8.3767	-0.4072	-7.9695	315.588	-81.944	3.683
3	-6.66	30.19	273.35	81.73	0.21	807.92	391.46	496.17	-8.4276	-0.3321	-8.0955	612.885	-115.165	4.773
4	-6.42	32.02	307.8	86.45	-0.02	856.18	426.84	529.12	-8.7675	-0.5477	-8.2198	662.881	-128.401	5.203
5	-10.47	36.86	323.41	102.93	0.74	929.03	428.72	555.43	-8.9754	-0.816	-8.1594	819.415	-134.982	3.017
6	-9.17	38.79	357.86	107.64	0.51	972.38	461.9	570.74	-8.8144	-0.7602	-8.0542	884.222	-148.212	2.042
7	-11.69	52.85	498.43	147.48	0.31	1267.03	562.93	699.84	-9	-0.793	-8.207	1509.675	-209.165	2.92
8	-10.51	54.78	532.87	152.2	0.09	1319.23	601.35	736.75	-8.9236	-0.8164	-8.1072	1568.075	-222.395	2.935
9	-12.9	40.92	362.45	114.22	-1.1	1002.12	431.25	582.03	-8.3744	-0.7691	-7.6053	1003.954	-152.477	3.268
10	-12.05	42.84	396.89	118.94	-1.32	1057.38	480	619.28	-8.3476	-0.6822	-7.6654	1052.557	-165.706	2.687
11	-10.67	47.99	433.53	132.87	-0.48	1168.2	499.06	635.99	-8.6678	-0.4303	-8.2375	1430.535	-188.471	3.068
12	-10.28	49.92	467.97	137.58	-0.7	1212.28	536.83	663.72	-8.3715	-0.5921	-7.7794	1505.713	-201.706	1.76
13	-12.17	39.33	353.44	109.21	0.32	1033.58	499.55	616.23	-8.5476	-0.8349	-7.7127	915.885	-152.476	2.96
14	-11.01	41.26	387.88	113.93	0.1	1069.17	522.53	640.25	-8.5086	-0.4504	-8.0582	963.045	-165.713	4.63
15	-25.85	38.18	377.42	102.14	1.08	1052.35	541.19	637.49	-8.7197	-0.5777	-8.142	1034.943	-175.713	3.713
16	-23.97	40.11	411.87	106.86	0.86	1091.79	562.02	656.7	-8.6615	-0.5768	-8.0847	1091.579	-188.949	2.642

H.E. = Hydrogen Energy(Kcal.mol⁻¹), Pol = Polarizability (A³), Mass = Molecular mass (amu), Ref = Refractivity (A³), Vol. = Volume (A³), S.A. = Surface area-Appro (A²), S.G = Surface Area-Grid (A²), HOMO = HOMO Energy level (eV), LUMO = LUMO Energy level (eV), E.gap = Energy gap (eV), N.E = Nuclear Energy (Hartree) T.E. = Total Energy (Hartree), D.M. = Dipole Moment (Debyes).

Eq. 1-4 gave correlation of parametric descriptors of thiazole derivatives with inhibitory zones of Pseudomonas areginosa. The number of descriptors is generally modified with an increase of correlation coefficient. It's clearly seen that Eq. 1 is influenced by four descriptors, Log p, HOMO, N.E., and T.E. In addition, the correlation coefficient value of this model is 0.806. While, Eq. 2 and Eq.3 depend on five descriptors which gives model with higher correlation coefficient, R²= 0.906 and 0.92,

respectively.

i.z. = -2.8419 Log p-3.4188 HOMO-0.0119 N.E-0.1117 T.E.-31.2879 (1)

n=16, $R^2=0.806$, F=11.477, S=0.873

i.z. = -0.5004 Pol.-4.9574 Log p+0.0453 S.A.-8.9095 HOMO-0.7735 D.M.-64.8516 (2)

n=16, R²=0.906, F=19.344, S=0.637

i.z. = -0.1627 Ref-4.9139 Log p+0.0397 S.A.-8.186 HOMO-0.798 D.M.-64.9613 (3)

$$n=16$$
, $R^2=0.92$, $F=23.019$, $S=0.589$

According to these equations, positive values of parameters refer to positive relationship, while negative parameters values refer to reversible relationship [23], Compared with Eq. 1-3, the robust model with correlation coefficient (R^2 =0.955) can be found in Eq. 4., and statistical data R², F and S were improved by using six descriptors, Ref, Log p, HOMO, N.E., T.E., and D.M. The statistical procedure known as regression basically involves drawing and analyzing trend-lines through data points, which are experimental and calculated data [24], as plotted in Figure.2. This model is validity to establish antibacterial activities for predicted molecules.

i.z. = -0.1053 Ref-4.1055 Log p-6.6947 HOMO-0.011 N.E.-0.1516 T.E.-0.6691 D.M.-53.0278 (4)

n=16, $R^2=0.955$, F=32.023, S=0.464

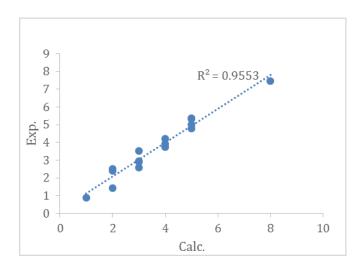


Figure 2: Experimental i.z. vs. Predicted i.z. Calculated By Eq. 4

Correlation of inhibitory zone of next gram

positive bacteria, Bacillus substilis, with thiazole parametric descriptors was also found and depicted in Eq. 5-8. The Eq. 5 needed five descriptors to build model with correlation coefficient value $R^2=0.817$. while, Eq. 6 and Eq. 7 needed six descriptors to build better models with regression coefficient, about $R^2 = 0.824$ $R^2 = 0.884$, respectively. Well-founded interrelationship model between antibacterial activity and other parameters goes to Eq. 8 with $R^2=0.915$. Good relationship between the experimental data and predicted antibacterial activities was illustrated as trend-line in Figure.3.

i.z. = 0.2419 H.E.-1.761 Log P-0.0263 N.E.-1.2039 T.E.-3.8021 D.M.+8.9435 (5)

n=16, R²=0.817, F=8.943, S=1.18

i.z. = 0.1986 Mass-0.3691 Ref-1.8126 Log p+0.0563 S.G.+0.339 T.E.-1.0465 D.M-5.9672 (6)

n=16, R²=0.824, F=7.024, S=1.08

i.z. = 0.2186 H.E-3.0612 Log P-4.622 E.gap-0.0313 N.E-0.2939 T.E.-1.6282 D.M.-40.77133 (7)

n=16, $R^2=0.887$, F=11.882, S=0.89

i.z. = 2.6578 Pol.+0.0734 Mass-0.7839 Ref-2.8401 Log P-4.626 E.gap-0.0296 N.E.-1.5777 D.M.- 45.7739 (8)

n=16, R²=0.915, F=12.408, S=0.82

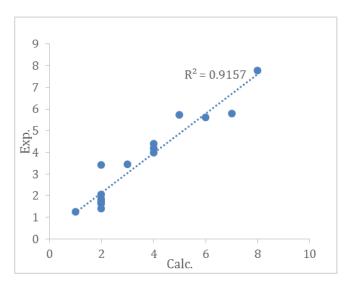


Figure 3: Experimental i.z. vs. Predicted i.z. Calculated By Eq. 8

On the other hand, the inhibitory zone of Staphylococcus aereus, gram negative bacteria, by thiazoles was analytically regressed to generate Eq. 9-12. Fairly good correlation coefficients were obtained with least number of descriptors, $R^2=0.855$, 0.889, 0.892 for Eq. 9, Eq. 10, and Eq. 11, respectively. Although of those good interrelationships, but parametric model was developed to get better results. Eq. 12 is the best predicated which depends on only 6 parameters, and it also provides good model with correlation coefficient, $R^2 = 0.954$. Figure.4 shows graph of experimental and predicted data obtained by Eq. 12.

i.z. = -1.5409 Log p-0.009 N.E-0.0927 T.E-0.6041 D.M.- 0.8647 (9)

n=16, R²=0.855, F=16.336, S=0.59

i.z. = 0.02803 Mass-1.4328 Log p-0.0351 Vol+0.5815 S.G.-0.5815 D.M.-1.262 (10) 10

n=16, R²=0.889, F=15.685, S=0.55

i.z. = 3.8833 Pol-0.8946 Ref-0.0925 Vol+0.0709 S.G.+3.346 E.gap+24.7004 (11)

n=16, $R^2=0.892$, F=16.586, S=0.53

i.z. = 2.9591 Pol-0.687 Ref-1.0827 Log p-0.086 Vol+0.087 S.G.-0.6595 D.M.-1.9898 (12)

n=16, R²=0.954, F=31.531, S=0.36

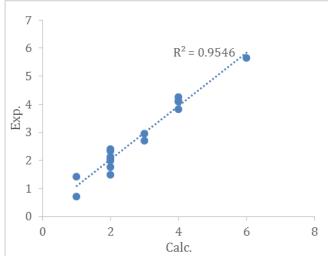


Figure 4: Experimental i.z. vs. Predicted i.z. Calculated By Eq. 12

Attempting to fit models correlate thialzole structure and inhibitory zone against Escherichia coli bacteria were statistically generated in Eq. 13-16. There is a significant relationship fitted on Eq. 16 mutually related with seven descriptors, and its correlation coefficient value was 0.918, which modeled and plotted as line-graph in Figure 5. Other models give the least correlation coefficient values, R²=0.899 for Eq. 15, R²=0.862 for Eq. 14, and R²=0.819 for Eq. 13.

i.z. = 0.4.5139 Pol-0.8528 Ref-0.1614 Vol+0.1459 S.G.-4.7032 (13)

n=16, R²=0.819, F=12.51, S=0.78

i.z. = -0.6688 H.E.+8.3923 Pol+0.267 Mass-2.492 Ref+0.9326 T.E-15.5742 (14) n=16, R²=0.862, F=12.558, S=0.71

i.z. = -0.2533 H.E.+6.843 Pol-1.4479 Ref-0.0735 Vol-0.0735 N.E.-9.7352 (15)

n=16, $R^2=0.899$, F=17.874, S=0.61

i.z. = -0.2195 H.E.+7.1124 Pol-1.5494 Ref-0.0708 Vol-0.0292 N.E.-0.1368 D.M.-9.33 (16)

n=16, $R^2=0.918$, F=16.95, S=0.58

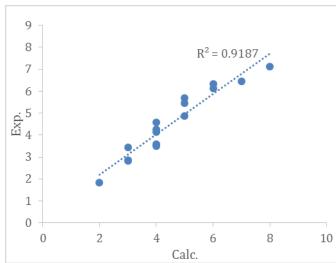


Figure 5: Experimental i.z. vs. Predicted i.z. Calculated By Eq. 16

Analysis of synthesized thiazole derivatives as antibacterial activities against Klebsiella pneumoniae bacteria was regressed. As a result, three first models represented in Eq. 17-19 have correlation coefficient values 0.884, 0.855, and 0.815, respectively. While, the last one, Eq. 20, shows excellent relationship to estimate activity that was achieved mathematically by parameters, and has good correlation coefficient value, R²=0.902. Best-fit line was created and drown in Figure 6 using both experimental and calculated variables of the last model.

i.z. = 0.1089 Mass-0.0935 Vol-0.0915

S.A.+0.1846 S.G.-7.7845 (17)

n=16, R²=0.815, F=12.116, S=1.13

i.z. = 0.0978 Mass-1.0218 Log p-0.0898 Vol-0.063 S.A.+0.1622 S.G.-7.5626 (18)

n=16, R²=0.855, F=6.719, S=1.05

i.z. = 0.0169 Mass-0.3762 Ref-0.1616 S.A.+0.1404 S.G.-0.0144 N.E.-0.7148 D.M.-7.4443 (19)

n=16, $R^2=0.8843$, F=11.47, S=0.99

i.z. = 0.1681 Mass-0.3461 Ref-0.1513 S.A.+0.1317 S.G.+1.584 HOMO-0.0155 N.E.-0.6542 D.M.+4.727 (20)

n=16, $R^2=0.902$ F=10.57, S=0.96

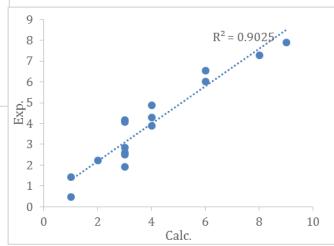


Figure 6: Experimental i.z. vs. Predicted i.z. Calculated By Eq. 20

A reliable equation, i.e. only high correlation coefficient (R²>9) equation, was used dependably to predict antibacterial activities of other 15 designed and unprepared conjugated thiazole-amino acids, numbered 17-31, as shown in Figure 7. Theoretical physio-chemical descriptors for proposed and untested molecules were determined and

listed in Table 3. Then, they were applied statistically in Eq. 4, Eq. 8, Eq. 12, Eq. 16, and Eq. 20 to calculate their inhibitory zones as predictable results. By observation of calculated results in table 4, it can be seen that compounds no. 28 and 29 show good antibacterial activity against Klebsiella pneumoniae, with 9.7 mm and 9.2 mm i.z., respectively. Two compounds, no. 18 and 29, can easily observe that higher inhibitory zones against Staphylococcus aereus, and five of the 15 designed molecules gave best antibacterial activities against Escherichia

coli, in particular, 22.6 mm i.z. by no. 18 and 22.7 i.z. mm by no. 29. Most of the predicted compounds exhibited good results against gram positive bacteria. For example, 13.5 mm i.z. of Bacillus substilis with compound no. 25 and 15.9 mm i.z. of Pseudomonas areginosa with compound no. 28.

Table 3: Calculated Physio-Chemical Descriptors of the Proposed Compounds

Molecule No.	H.E.	Pol.	Mass	Ref.	Log p	Vol.	S.A.	S.G.	номо	LUMO	E. gap	N.E.	T.E.	D.M
17	-11.18	27.29	267.73	74.39	-0.95	732.52	387.99	461.7	-8.6424	-0.5321	-8.1103	496.54	-112.26	1.932
18	-9.72	29.13	281.76	68.21	2.17	779.26	413.14	481.98	-8.6669	-0.5231	-8.1438	555.44	-117.98	2.043
19	-8.09	32.8	309.81	87.88	0.46	864.84	446.52	521.8	-8.6237	-0.487	-8.1367	686.47	-129.43	3.429
20	-7.67	34.63	323.84	92.56	0.79	921.65	491.8	558.99	-8.6561	-0.5089	-8.1472	749.00	-135.15	3.151
21	-7.53	34.63	323.84	92.48	0.86	913.45	476.34	546.43	-8.6222	-0.4838	-8.1384	751.13	-135.15	3.469
22	-13.92	29.76	297.76	80.43	-1.19	800.49	408.24	496.19	-8.5531	-0.4671	-8.086	620.24	-129.77	3.234
23	-12.31	31.6	311.79	84.85	-0.78	838.69	421	514.33	-8.5114	-0.4865	-8.0249	689.72	-135.49	3.331
24	-10.86	32.13	313.82	86.55	-0.61	824.72	425.34	510.87	-8.5561	-0.5375	-8.0186	614.74	-125.12	3.929
25	-9.44	35.8	341.87	96.22	-0.42	937.17	505.49	570.7	-8.6981	-0.6033	-8.0948	737.20	-136.57	1.486
26	-13.86	35.98	338.85	96.2	-0.64	972.52	509	590.02	-8.6951	-0.5803	-8.1148	803.37	-143.27	1.803
27	-21.01	38.27	366.87	101.89	-0.81	1028.22	502.8	622.91	-8.5471	-0.4201	-8.127	925.77	-158.45	5.008
28	-12.5	34.12	333.8	93.22	-2.22	864.4	372.11	511.42	-8.9827	-1.0346	-7.9481	803.21	-142.54	4.897
29	-16.58	31.69	325.77	74.25	1.55	847.62	436.28	525.88	-8.5396	-0.5194	-8.0202	731.12	-146.26	4.595
30	-15.29	33.52	339.8	89.67	-0.78	909.6	481.08	558.28	-8.7248	-0.6027	-8.1221	783.53	-151.98	2.465
31	-14.99	32.4	324.78	86.74	-1.89	863.23	428.84	531.44	-8.5166	-0.4188	-8.0978	731.69	-142.59	6.274

Figure 7: Chemical Structures of Proposed Thiazole Derivatives

Table 4: List Of Predicted Inhibitory Zones (i.z.) of Proposed Thiazole Derivatives Against Five Type of Bacteria.

Calculated inhibitory zone diameter (mm) Gram negative bacteria Gram positive bacteria Klebsiella Staphylococcus Escherichia Bacillus Pseudomonas Molecular Molecular name pneumoniae aereus coli substilis areginosa No. 17 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)acetamide 5.3 3.4 4.6 10.6 11.2 18 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)propanamide 5.8 22.6 10.7 8.6 -0.719 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-3-3.0 7.8 5.9 1.0 3.3 methylbutanamide 20 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-4--0.93.3 7.7 5.8 2.1 methylpentanamide 21 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-3--0.5 2.7 8.3 5.0 1.3 methylpentanamide 22 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-3-5.2 4.3 5.5 9.5 11.3 hydroxypropanamide 23 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-3-5.5 4.5 6.7 8.3 8.9 hydroxybutanamide 24 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-3-4.8 5.2 10.6 9.2 7.2 mercaptopropanamide 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-4-25 1.4 6.4 10.2 8.4 13.5 (methylthio)butanamide 2.6-diamino-N-(4-(4-chlorophenyl)thiazol-2-26 1.7 5.6 8.0 12.0 9.4 yl)hexanamide 27 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-5-5.9 4.6 9.1 7.6 7.3 guanidinopentanamide 2-amino-N-(4-(4-chlorophenyl)thiazol-2-yl)-2-(1H-28 9.7 4.3 6.3 7.9 15.9 imidazol-5-yl)acetamide 3-amino-4-((4-(4-chlorophenyl)thiazol-2-yl)amino)-4-29 9.2 8.9 22.7 8.0 1.0 oxobutanoic acid 30 4-amino-5-((4-(4-chlorophenyl)thiazol-2-yl)amino)-5-4.0 5.2 5.9 10.7 11.9 oxopentanoic acid 2-amino- N^1 -(4-(4-chlorophenyl)thiazol-2-yl)succinamide 31 5.5 4.2 6.7 7.5 12.0

5. CONCLUSION

The models of QSAR study of 16 derivatives amino acids conjugated 2-amnioarylthiazole were successfully investigated as in vitro anti-bacteria. By using MLR method with help of the numerous descriptors, it can correlate with experimental inhibitory zones of bacteria. Five excellent models with high correlation coefficient, high fisher ratio, and low standard error have been built. QSAR studies indicated that strong relationship between (Ref., Log p, HOMO, N.E, T.E., and D.M.) zone of Pseudomonas and inhibitory

areginosa was found in Eq. 4 with $R^2=0.955$. Also, Eq. 8 with $R^2=0.915$ was found to contribute to antibacterial activity against substilis which Bacillus depend parameters (Pol., mass, Ref., Log P, E. gap, N.E., and D.M.). Further, (Pol., Ref. Log p. Vol., S.G., and D.M.) were the key properties for explaining thiazole anti-Staphylococcus aereus activity which was achieved in Eq. 12 with R²=0.954. Statistical significance in the screening best equation to calculate predictably anti-Escherichia coli was directly influenced on descriptors (H.E., Pol., Ref, Vol., N.E., and D.M.), and

founded in Eq. 16 with R²=0.918. In addition, anti-Staphylococcus aereus was well linked with parameters (Mass., Ref., S.A., S.G., Homo, N.E. and D.M.) in Eq. 20 with R²=0.902. Proposing more thiazole derivatives gives subset of variables are more effective as anti-bacteria. Good results of new untested-set evaluated at Eq. 4, Eq. 8, Eq.12, Eq.16, and Eq. 20, which showed most of them good inhibitors for gram positive of bacteria.

REFERNCES

- [1] C. Nantasenamat, C. Isarankura-Na-Ayudhya, T. Naenna and V. Prachayasittikul, "A practical overview of quantitative structure-activity relationship", *EXCLI Journal*, Vol. 8, 2009, pp. 74-88.
- [2] C. Hansch, B. R. Telzer and L. Zhang, "Comparative **QSAR** in toxicology: examples from teratology and cancer chemotherapy of aniline mustards", Critical Rev Toxicol, Vol. 25, No. 1, 1995, pp. 67-89. [3] W. Tong, R. Perkins, R. Strelitz, E. R. Coliantes, S. Keenan, W. J. Welsh, W. S. Branham, and D. M. Sheehan, "Quantitative Structure-Activity Relationships (QSARs) for Estrogen Binding to the Estrogen Receptor: Predictions across Species", Environmental Health Perspectives, Vol. 105, No. 10, 1997, pp. 1116-1124.
- [4] J. H. Al-Fahemi, D. L. Cooper, and N. L. Allan, "Predictions of Toxicity to Chlorella vulgaris and the Use of Momentum-space Descriptors", *Croatica Chemica Acta*, Vol. 82, No. 1, 2009, pp. 311-316.
- [5] E. U. Ramos, W. H. Vaes, H. J. Verhaar, and J. L. Hermens, "Polar narcosis: Designing a suitable training set for QSAR Studies", *Environmental Science and Pollution Research*, Vol. 4, No. 2, 1997, pp. 83-90.
- [6] W. Shao-peng, J. Zhi-qin, Z. Hui-xiao, Z. Ji-wen, W. Yong-hua, and W. Wen-jun, "Isolation, biological evaluation and 3D-

- QSAR studies of insecticidal/narcotic sesquiterpene polyol esters", *Journal of Molecular Modeling*, Vol. 17, 2001, pp. 681-693.
- [7] K. A. Hussain, W. A-H Radhiand, and S. M-H Ismael, "Quantitative Structure-Activity Relationships (QSAR) study and improving it of some schiff-base ligands as anticancer for prostate cancer", *Journal of Chemical and Pharmaceutical Research*, Vol. 4, No. 3, 2012, pp. 1702-1707.
- [8] F. A. Alvarez Núñez, and S. H. Yalkowsky, "Correlation between log P and ClogP for Some Steroids", *Journal of Pharmaceutical sciences*, Vol. 86, No. 10, 1997, pp. 1187-1189.
- [9] L. Figueroa-Valverde, F. Díaz-Cedillo, M. López-Ramos, and E. Garcia-Cervera, "A facile Synthesis of two quaternary aminosteroids derivatives and its correlation with descriptors LogP and π ", *International Journal of ChemTech Research*, Vol. 2, No. 3, 2010, pp. 553-1559.
- [10] Y. Li, Y. Wang, L Yang, S. Zhang, and C. Liu, "Impact of Molecular Hydrophobic Field on Passive Diffusion, P–Glycoprotein Active Efflux, and P–Glycoprotein Modulation of Steroids", *Internet Electronic Journal of Molecular Design*, Vol. 5, No. 2, 2006, pp. 60-78.
- [11] P. Bhattacharya, J. T. Leonard, and K. Roy, "Exploring QSAR of thiazole and thaidiazole as potent and selective human adenosine A3 receptor antagonists using FA and GFA techniques", *Bioorganic & medicinal Chemistry*, Vol 13, No. 4, 2005, pp. 1159-1165.
- [12] A. Borghini, D. Pietra, P. Domenichelli, and A. M. Bianucci, "QSAR study on thiazole and thiadiazole analogues as antagonists for the the adenosine A1 and A3 receptors", *Bioorganic & medicinal Chemistry*, Vol. 13, No. 18, 2005 pp. 5330-5337.
- [13] P. K. Balasubramanian, A. Balupuri,

- and S. J. Cho, "Molecular modeling studies of trisubstituted thiazoles as Cdc7 kinase inhibitors through 3D-QSAR and molecular docking simulation", *Bulletin of the Korean Chemical Society*, Vol. 36, No. 6, 2015, pp. 1599-1612.
- [14] B. Elzbieta, S. Justyna, and S. Anna, "A structure-activity relationship study of thiazole derivatives with H₁-antihistamine activity", *Acta Poloniae Pharmaceutica-Drug Research*, Vo. 68, No. 5, 2011, pp. 677-686.
- [15] A. Muala and K. Ali, "Theoretical Semi-Empirical Calculation of Molecular Structure of Schiff Base Complexes Zn (II)", *Jouranl of Thi-Qar Science*, Vol 2, No. 4, 2011, pp. 96-108.
- [16] S. M. Ismael, "QSPR model of Intrinsic Viscosity for Poly(isobutylene)", *Journal of Computational Methods in Molecular Design*, Vol. 2, No. 4, 2012, pp.130-135.
- [17] P. Arora, R. Narang, S. Bhatia, S. Kumar Nayak, S. K. Singh, and B. Narasimhan, "Synthesis, molecular docking and QSAR studies of 2, 4-disubstituted thiazoles as antimicrobial agents", *Journal of Applied Pharmaceutical Science*, Vol. 5, No. 20, 2015, pp. 28-42.
- [18] G. A. Kashid, J. Sarvanan, and N. P. Jain, "QSAR studies on thiazoles having antiplatelet activity", *Innovational Journal of Chemistry*, Vol 1, No. 1, 2015, pp. 10-21.
- [19] K. C. Prakasha, G. M. Raghavedra, R. Harisha, and C. Gowda, "Design, synthesis and antimicrobial screening of amino acids conjugated 2-amnio-arylthiazole derivatives" *International Journal of Pharmacy and Pharmaceutical Science*, Vol 3, No. 3, 2011, pp. 120-125.
- [20] C. Prez, M. Paul, and P. Bazerque, "An antibiotic assay by the agar well diffusion method", *Acta Biologiae et Medicine Experimentalis*, Vol. 15, 1990, pp. 113-115. [21] R. B. Darlington, "Regression and Linear Models", 1990, McGraw-Hill, New

York.

- [22] D. L. Massart, B. G. Vandeginste, L. M. Buydens, S. D. Jong, P. J. Lewi, and J. Smeyers-Verbeke, "Handbook of Chemometrics and Qualimetrics: Part A", 1997, Elsevier, Amsterdam.
- [23] S. M. Ismael, K. A. Hussain, W. A. Radhi, and H. A. Majeed, "Quantitative structure property relationship (QSPR) study of phthalate plasticization for PVC", *Wasit Journal for Science & Medicine*, Vol. 7, No. 4, 2014, pp. 225-233.
- [24] J. Guerard. and B. John," Introduction to Financial Forecasting in Investment Analysis", 2013, Springer Science, New York.

دراسات علاقة التركيب الفعالية الكمية للأحماض الأمينية المقترنة مع مركب ٢ -امينوارايل ثايوزول كمضاد للبكتريا

ماهر اسعد جلال ، وسام عبد الحسن راضي جامعة البصرة – مركز أبحاث البوليمرقسم الكيمياء وتكنولوجيا البوليمرات

المستخلص

تم في هذا البحث دراسة علاقات التركيبية الفعالية الكمية (QSAR) الى ستة عشر مشتق من مركبات الثايوزول كمضاد بكتريا (Klebsiella pneumonia) و Klebsiella pneumonia و Bacillus substilis وcoli و Bacillus substilis ودoli و Bacillus substilis الفيز المستغدام طريقة ميكانيك الكم (MM+) وكذلك طريقه كيمياوية حسابياً للتراكيب الكيميائية لمشتقات الثايوزول باستخدام طريقة ميكانيك الكم (MM+) وكذلك طريقه شبه التجريبية (AM₁) عند الطاقة الدنيا للمركبات ، ثم أنجزت المعادلات التي تربط الصفات التركيبية للمشتقات الثايوزول مع فعاليتها كمضاد للبكتريا باستخدام التحليل الخطي المتعدد الارتداد ، حيث تم استنتاج ۲۰ موديل يتراوح معاملها الارتباط (P2>0.9) واستخدمت بعدها اعلى قيمة ارتباط (P2>0.9) للتنبؤ بفعالية كمضاد للبكتريا الى خمسة عشر مركب مفترض من الاحماض الامينية المقترنة مع ۲-امينوارايل ثايازول ، واظهرت النتائج ان معظم المركبات المفترضة لها فعالية جيده ضد بكتريا كرام الموجبة.