

Synthesis ,Characterization and Identification of New Oxazepine,Oxazepane and Oxazinan Compounds from reaction of Some (Z)-2-((2-hydroxybenzylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol Derivatives with some cycloanhydrides.



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ABSTRACT

(Z)-2- ((2-hydroxybenzylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol, (E)-4-(3-hydroxy-2,2-bis(hydroxymethyl)propylideneamino)pentan-2-one, -2-((4-hydrazonopentane-2-ylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol and (Z)-2-((4-amino-1,5-dimethyl-2-phenyl-1H-pyrazol-3(2H)-ylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol were prepared by condensation of 2-(aminomethyl)-2-(hydroxymethyl)propane-1,3-diol with 2-hydroxy benzaldehyde, pentane-2,4-dione, (E)-4-hydrazonopentan-2-one and 4-amino-1,5-dimethyl-2-phenyl-1H-pyrazol-3(2H)-one. These Schiff-bases were reacted with Succinic, Maleic and Malonic anhydride in absolute ethanol to give 6,7-membered heterocyclic ring system Oxazepine, Oxazepane, Oxazinan. The Schiff-bases and the final products were identified by their melting points, elemental analysis, FT-IR and UV-Visible spectra.

Introduction

The synthesis of 2-phenyl-1,3-oxazepine by irradiation of 4-phenyl-2-oxa-3-aza bicyclo[3.2.0]hepta-3,6-diene was studied by Toshio Makai with other workers [1]. The discovery of the central nervous system (CNS) activity of 1,4-benzodiazepine [2] encourage the chemists to look for more effective ways to build up the 7-membered heterocyclic ring systems from already available materials. The Six-membered heterocyclic ring system: 1,3-oxazine has already been reported and thoroughly reviewed in the literature [1-4]. Maleic, arylmaleic and substituted maleic anhydrides react with trimethylsilyl azide to give 4- and 5-substituted ((oxauraciles)) dihydro-1,3-oxazine-2,6-diones [5,6].

Both 2-methoxy pyrroline and 2-methoxy piperidine react with diketene under neutral conditions at 0°C to give the corresponding, 2-methoxy dihydro 1,3-oxazine-4-ones [7,8].

One of these ways which has been discovered recently involves direct addition of maleic anhydride to the (C=N) double bond of Schiff-bases and

a number of 2,3-diaryl-2,3-dihydro-1,3-oxazepine-4,7-diones were prepared and characterized. [5-8]

Diketene reacts with N,N-diphenyl guanidine to give the tranquilizer ketazolam and N-substituted tetrahydro-1,3-oxazine-4-one respectively [9,10]. The reaction of diketene with isocyanic acid, cyanamides and fluoro sulphonyl isocyanate afforded the corresponding 1,3-oxazine-2,4-diones [11-13].

Furthermore, thermal rearrangement of ketovinylazirines gave substituted 1,3-oxazepines. [9-14]

N-acyl immonium ions have been the most commonly used dienes to effect [4+2] cycloaddition as 4π components with substituted 1,3-butadienes. It is found that N-acylimines or immonium ions that are capable of tautomerization undergo intermolecular Diels-alder reaction to give dihydro-1,3-oxazines [17].

The reaction of N-Benzylidene 1,5-dimethyl-2-phenylpyrazolonamines (Schiff bases with

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Cyclopentane -1,1-dicarboxylic anhydride to give 2-aryl-3-3(1,5-dimethyl-2-phenylpyrazolo)-1-(5) spirocyclopentyltetra hydro-1,3-oxazine-4,6-diones [18]. Synthesis and characterization of 8-(4-dimethyl amino-phenyl)-9-(6-R- benzothiazol-2-yl)-7-oxa-9-aza-spiro[4.5]decane-6,10-dione[19].

Extensive synthesis and testing of the barbiturates over a long time span has produced well-defined structure – activity relationships. Which have been summarized. [20] .

Materials and methods

Melting points were recorded with Gallenkamp Melting point Apparatus and were uncorrected. Elemental analysis were carried out with perkin-Elmer,2400;CHN Elemental Analyzer. FT-IR spectra were recorded on FT-IR spectrophotometer -8400s Shimadza (KBr) in Chemistry department of Education for women college AL-Anbar and UV-Visible spectra were recorded (in ethanol) on Schimadza Reco- 160 Spectrophotometer.

Preparation of (Z)-2-((2-hydroxybenzylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol:

A mixture of 0.01 mole(1.35 g) of 2-(amino methyl)-2-(hydroxymethyl)propane-1,3-diol and 0.01 mole (1.22 g) of o-hydroxy benzaldehyde in 20 ml of absolute ethanol, was refluxed in a water bath for 1h, then left to cool in ice-water bath whereby yellow crystalline solid was precipitated. The solid was filtered, recrystallized twice from ethanol.

Preparation of (E)-4-(3-hydroxy-2,2-bis(hydroxymethyl)propan-2-one:

A mixture of 0.01 mole(1.35 g) of 2-(amino methyl)-2-(hydroxymethyl)propane-1,3-diol and 0.01 mole (1.0 g) of acetyl acetone in 20 ml of absolute ethanol,

was refluxed in a water bath for 1h, then left to cool in ice-water bath whereby yellow crystalline solid was precipitated. The solid was filtered, recrystallized twice from ethanol.

Preparation of 2-((4-hydrazonopentan-2-ylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol:

A mixture of 0.01 mole(2.17 g) of (E)-4-(3-hydroxy-2,2-bis(hydroxymethyl)propan-2-one and 0.01 mole (0.5 g) of hydrazine hydrate in 20 ml of absolute ethanol, was refluxed in a water bath for 1.5h, then left to cool in ice-water bath whereby yellow crystalline solid was precipitated. The solid was filtered, recrystallized twice from ethanol.

Preparation of(E)-4-((3-hydroxy-2,2-bis(hydroxymethyl)propylimino)methyl)-5-methyl-2phenyl-1H-pyrazol-3(2H)-one :

A mixture of 0.01 mole(1.35)of 2-(amino methyl)-2-(hydroxymethyl)propane-1,3-diol and 0.01 mole (2.03g)of 4-aminoanti pyrine in 20 ml of absolute ethanol, was refluxed in a water bath for 1h, then left to cool in ice-water bath whereby pink crystalline solid was precipitated. The solid was filtered, recrystallized twice from ethanol.

Preparation of 3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-(2-hydroxyphenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione:

In a 100 ml round bottom flask equipped with a double surface condenser fitted with calcium chloride guard tube, was placed a mixture of 0.01mol (2.39g) of (Z)-2-((2-hydroxybenzylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol and 0.01 mol (0.98g)of maleic anhydride in 20ml of absolute ethanol.

The reaction mixture was refluxed in water bath at 78°C for 2hr., the solvent was then removed and the resulting solid was recrystallized from anhydrous THF. Compounds 8,11,14 preparation in same of method.

Preparation of 3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-(2-hydroxyphenyl)-1,3-oxazepane-4,7-dione:

In a 100 ml round bottom flask equipped with a double surface condenser fitted with calcium chloride guard tube, was placed a mixture of 0.01mol (2.39g) of (Z)-2-((2-hydroxybenzylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol and 0.01 mol (1.0g) of succin anhydride in 20ml of absolute ethanol. The reaction mixture was refluxed in water bath at 78°C for 2hr., the solvent was then removed and the resulting solid was recrystallized from anhydrous THF. Compounds 9,12,15 preparation in same of method.

Preparation of (R)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-2-(2-oxopropyl)-1,3-oxazinane-4,6-dione:

In a 100 ml round bottom flask equipped with a double surface condenser fitted with calcium chloride guard tube, was placed a mixture of 0.01mol (1.35g) (E)-4-(3-hydroxy-2,2-bis(hydroxymethyl)propylimino)pentan-2-one 0.01 mol (0.86g) of malonic anhydride in 20ml of absolute ethanol. The reaction mixture was refluxed in water bath at 78°C for 3hr., the solvent was then removed and the resulting solid was recrystallized from anhydrous 1,4-Dioxan. Compounds 10,13,16 preparation in same of method.

Results and Discussion

Schiff bases (1-4) are prepared by condensation of 2-(amino methyl)-2-(hydroxymethyl)propane-1,3-diol with salicylaldehyde, acetyl acetone and 4-amino antipyrine .

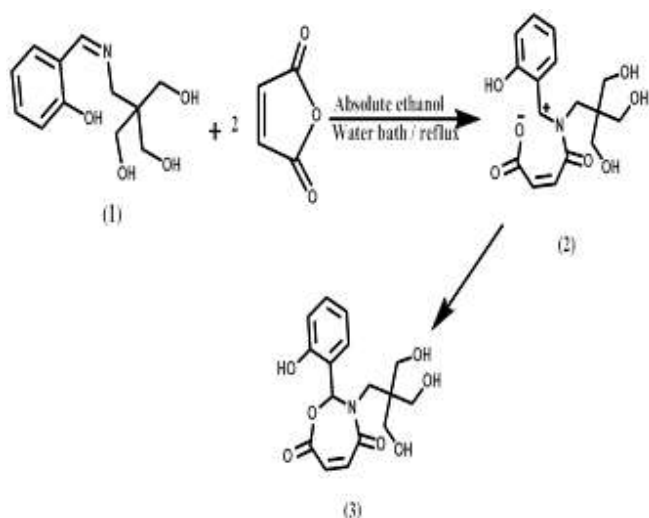
It is known that Schiff bases react smoothly with acid chlorides and anhydrides to give the corresponding addition products[21-24].

This is indicated by the appearance of the characteristic C=O (lacton-lactam) absorption band at (1665-1680)cm⁻¹ in the FT-IR spectra of addition products (3).

It is impressive to note that the two absorption band at (1800-1950)cm⁻¹ in the FT-IR spectra of pure Maleic ,Succinic and Malonic anhydride have disappeared when the anhydride became of the 6,7-membered ring system of the 3-(3-hydroxy-2,2-bis(hydroxymethyl) propyl)-2-phenyl-1,3-oxazepane-4,7-dione, 3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-phenyl-1,3-oxazepine-4,7-dione and 2-(4-(dimethylamino) phenyl)-3-(3-hydroxy-2,2-bis(hydroxymethyl) propyl)-1,3-oxazinane -4,6-dione. The new absorption bands of the (C=O) group in the FT-IR spectra of the addition products (3) appear at (1665-1685)cm⁻¹ , this attributed to the fact that the structures of the addition products are combination of the lacton-lactam structure [25]

Furthermore, the UV-Visible spectra of Oxazepine, Oxazinane derivatives show absorption maxima at (240-350)nm due to charge transfer of the cyclic 6,7-membered lactone-lactam combined structure [3]. and positive Br₂/CCl₄ and KMNO₄ tests.

In this paper , the reaction of the Maleic, Succinic and Malonic anhydride with (1-4) Schiff bases gives the dipolar intermediate [2] which collapses to 6,7-membered heterocyclic ring system. [3] is presented.



Cycloaddition is a ring formation that results from the addition of bonds to either δ or π with formation of new δ bonds. This class of reactions and its reverse encompasses a large number of individual types. Huisgen (20) has formulated a useful classification of diverse cycloaddition in terms of the number of the new δ bond. The ring size of the product, and the number of atoms in the components taking part in the cycloaddition. This cycloaddition reaction is classified as a 2+5-7, and it is the first cycloaddition of this type, although in principle, one would predict that the butadiene cation might add to an olefin through a $(4n+2)$ transition state to yield the cyclohexenyl cation.

The reaction of maleic, succinic and malonic anhydride with various Schiff bases is a sort of cycloaddition reaction.

Table (1) Melting points, yield, Molecular formula [M.F], elemental analysis of New Schiff bases (1-4)

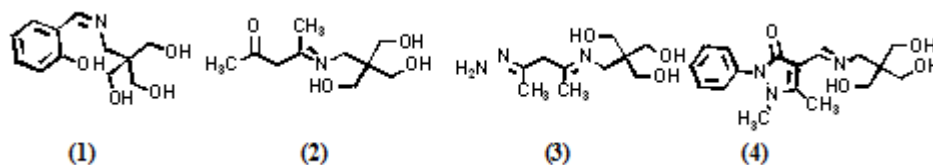


Table (2) The major FT-IR absorptions (cm⁻¹) of New Schiff bases (1-4)

Compound	O-H str.	N-H str.	C-H str. Aromatic	C=O str.	C=N str.	C=C str. Aromatic	C-O str.	C-H bend. Aromatic
1	3540	-	3050	-	1610	1580	1310	1010,770
2	3460	-	3070	1690	1605	1585	1320	1020,870
3	3510	3450,3310	3090	-	1620	1580	1310	1010,900
4	3470	3430,3280	3065	1680	1610	1590	1330	1020,870

Table (3) The UV-Visible absorption maxima λ /nm of New Schiff bases (1-4)

Compound	UV-Visible absorption maxima λ /nm
1	380,300,266,225,220
2	275,226,220
3	280,244,222
4	375,320,260,251,226

Table (4) Melting point, percentage yield, molecular formula and elemental analysis of compounds (5-16)

Comp	m.p/c	Yield%	M.F	Calc.			Found		
				C	H	N	C	H	N
5	177-175	72	C ₁₆ H ₁₉ NO ₇	56.97	5.68	4.15	56.68	5.55	4.01
6	190-188	66	C ₁₆ H ₂₁ NO ₇	56.63	6.24	4.13	56.45	6.18	3.98
7	211-209	68	C ₁₅ H ₁₉ NO ₇	55.38	5.89	4.31	55.12	5.80	4.21
8	240-238	65	C ₁₄ H ₂₁ NO ₇	53.33	6.71	4.44	53.19	6.65	4.20
9	166-164	80	C ₁₄ H ₂₃ NO ₇	52.99	7.31	4.41	53.04	7.16	4.23
10	170-168	71	C ₁₃ H ₂₁ NO ₇	51.48	6.98	4.62	51.29	6.80	4.54
11	246-244	67	C ₁₈ H ₂₅ N ₃ O ₉	50.58	5.90	9.83	50.41	5.74	9.70
12	188-186	70	C ₁₈ H ₂₉ N ₃ O ₉	50.11	6.78	9.74	50.02	6.51	9.48
13	165-163	66	C ₁₆ H ₂₅ N ₃ O ₉	47.64	6.25	10.42	47.46	6.23	10.28
14	154-152	67	C ₂₀ H ₂₆ N ₄ O ₆	57.41	6.26	13.39	57.31	5.99	13.10
15	206-104	59	C ₂₀ H ₂₈ N ₄ O ₆	57.13	6.71	13.33	57.03	6.52	13.08
16	258-256	57	C ₁₉ H ₂₆ N ₄ O ₆	56.15	6.45	13.79	56.01	6.28	13.62

Table(5)The major IR absorption (cm-1)of compounds (5-16)

Comp.	O-H str.	NH ₂ str.	C=O str.	C=O str. Lacton, lactam	C=N str.	C=C str. Olefin	C=C str. Aromatic	C-O-C str.
5	3510,3470	-	-	1680	1430	1600	1580,1560	1190
6	3520,3480	-	-	1680	1435	-	1580,1540	1180
7	3540,3470	-	-	1670	1450	-	1580,1560	1170
8	3570	-	1710	1685	1440	1610	-	1200
9	3560	-	1700	1670	1430	-	-	1210
10	3555	-	1705	1675	1450	-	-	1200
11	3530	3410,3360	-	1670	1480	1610	-	1190
12	3560	3430,3320	-	1665	1460	-	-	1210
13	3540	3430,3290	-	1680	1450	-	-	1200
14	3550	3420,3280	-	1670	1430	1600	1580,1555	1190
15	3560	3415,3275	-	1670	1460	-	1580,1565	1180
16	3540	3450,3300	-	1675	1450	-	1580,1565	119

Table (6) Uv-spectra of compounds (5-16)

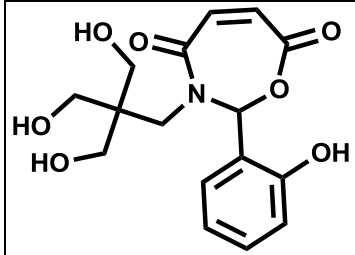
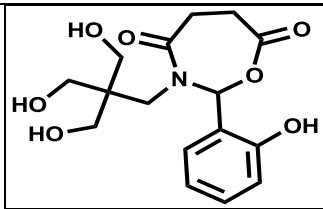
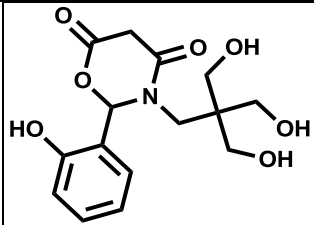
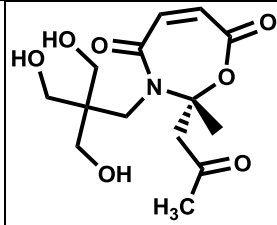
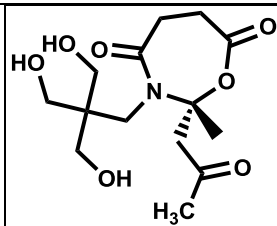
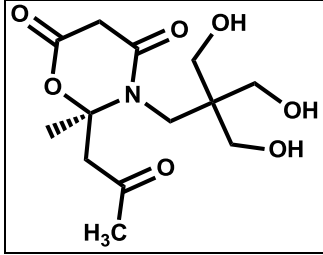
Comp.	UV-Visible absorption maxima λ/nm
5	320,300,266,230,221
6	315,255,243,229
7	333,265,251,243,223
8	285,265,239,224
9	289,269,241,236,222
10	285,240,225,220
11	275,250,235,226
12	295,250,235,222
13	290,269,240,235,220
14	310,300,276,220.
15	325,235,223
16	330,285,261,243,223

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No.	Name of compounds	Structure
1	(Z)-2-((2-hydroxybenzylideneamino) methyl)-2-(hydroxymethyl) propane-1,3-diol	
2	(E)-4-(3-hydroxy-2,2-bis(hydroxymethyl)propylimino)pentan-2-one	
3	2-((4-hydrazonopentan-2-ylideneamino) methyl)-2-(hydroxymethyl)propane-1,3-diol	
4	(Z)-2-((4-amino-1,5-dimethyl-2-phenyl-1H-pyrazol-3(2H)-ylideneamino)methyl)-2-(hydroxymethyl)propane-1,3-diol	

5	<p>3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-(2-hydroxyphenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione</p>	
6	<p>3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-(2-hydroxyphenyl)-1,3-oxazepane-4,7-dione</p>	
7	<p>3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-(2-hydroxyphenyl)-1,3-oxazinane-4,6-dione</p>	
8	<p>(R)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-2-(2-oxopropyl)-2,3-dihydro-1,3-oxazepine-4,7-dione</p>	
9	<p>(R)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-2-(2-oxopropyl)-1,3-oxazepane-4,7-dione</p>	
10	<p>(R)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-2-(2-oxopropyl)-1,3-oxazinane-4,6-dione</p>	
11		

	<p>(R)-3-amino-2-(((S)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-4,7-dioxo-2,3,4,7-tetrahydro-1,3-oxazepin-2-yl)methyl)-2-methyl-2,3-dihydro-1,3-oxazepine-4,7-dione</p>	
12	<p>(R)-3-amino-2-(((S)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-4,7-dioxo-1,3-oxazepan-2-yl)methyl)-2-methyl-1,3-oxazepane-4,7-dione</p>	
13	<p>(R)-3-amino-2-(((S)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2-methyl-4,6-dioxo-1,3-oxazinan-2-yl)methyl)-2-methyl-1,3-oxazinane-4,6-dione</p>	
14	<p>2-(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1H-pyrazol-4-yl)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-2,3-dihydro-1,3-oxazepine-4,7-dione</p>	
15	<p>2-(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1H-pyrazol-4-yl)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-1,3-oxazepane-4,7-dione</p>	
16	<p>2-(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1H-pyrazol-4-yl)-3-(3-hydroxy-2,2-bis(hydroxymethyl)propyl)-1,3-oxazepane-4,7-dione</p>	

تحضير وتشخيص ودراسة الصفات الفيزيائية لمركبات الأوكسازين، الأوكسازبان وألأوكسينان من 1-, 3, (هيدروكسي) بروبان-2- هيدروكسي بنزليدين أمينو) مثيل)-2- (Z)-تفاعل قواعد شيف - دايلول مع بعض الانهيدريدات الحلقية

وليد فرج حمادي خالد فاروق عبد الغفور حميد مدلول الكبيسي

الخلاصة :-

تعطي تفاعلات قواعد شيف جديدة (Z)-2- ((2-هيدروكسي بنزليدين أمينو) مثيل)-2- (هيدروكسي) بروبان-1-, 3- دايلول، (Z)-2-))
2-هيدروكسي بنزليدين أمينو) مثيل)-2- (هيدروكسي) بروبان-1-, 3- دايلول و (Z)-2-((4-ثنائي مثيل امينو (بنزليدين أمينو) مثيل)-
2- (هيدروكسي) بروبان-1-, 3- دايلول مع انهيدريدات السكسنيك ، المالك والمالونيك الحلقية مركبات سداسية و سباعية الحلقة ، الأوكسازينانات ،
الأوكسازينانات،الأوكسازينانات. تم التثبت من صحة التراكيب المحضرة من خلال تعيين درجات الانصهار و طيف الأشعة فوق البنفسجية - المرئية
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